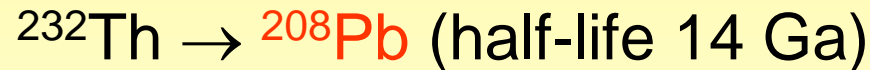
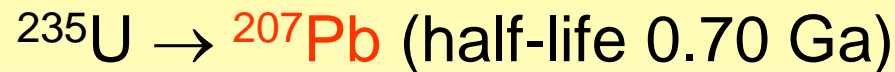
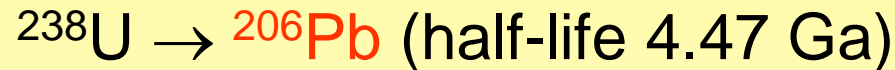


Die Blei-Blei Methoden

Pb produced by radioactive decay of U & Th

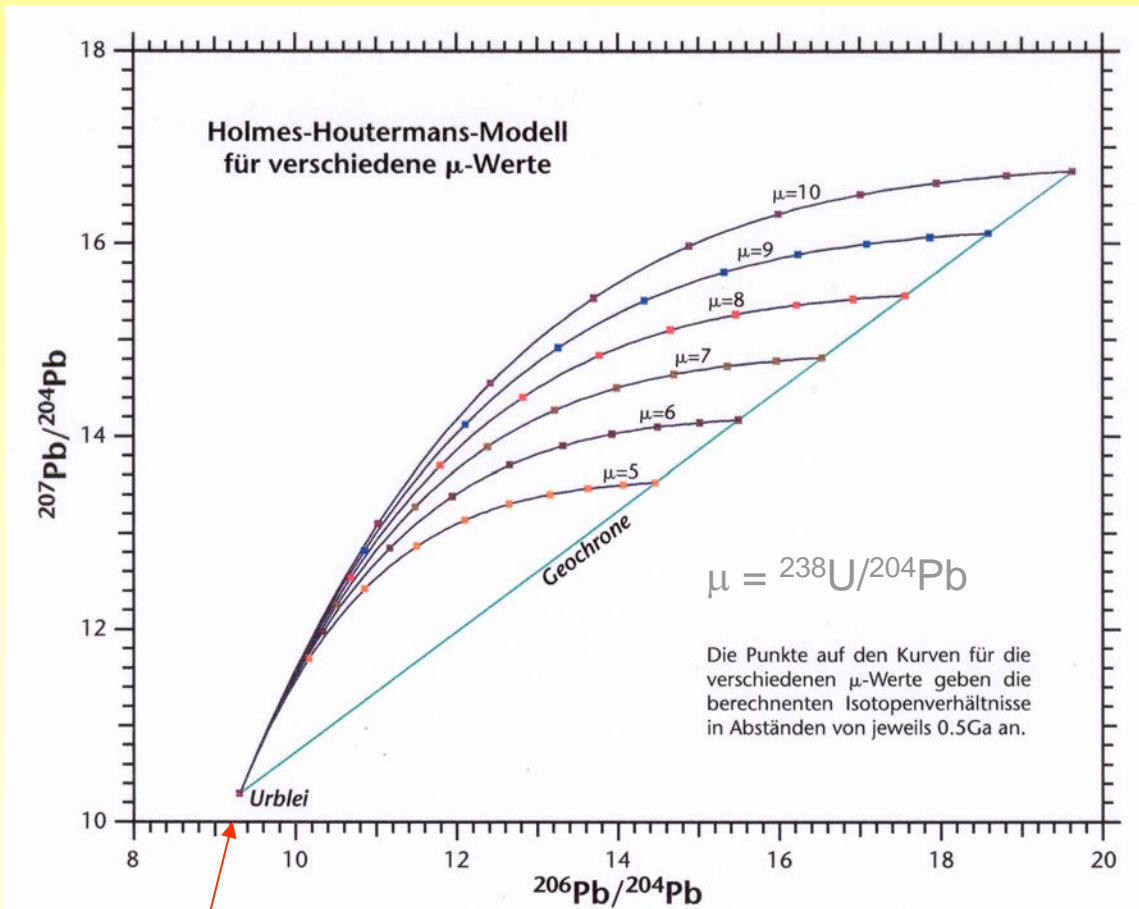


non-radiogenic ^{204}Pb used as reference

so, increase of $^{208}\text{Pb}/^{204}\text{Pb}$, $^{207}\text{Pb}/^{204}\text{Pb}$,
due to U and Th decay

$$\frac{\Delta ^{207}\text{Pb}/^{204}\text{Pb}}{\Delta ^{206}\text{Pb}/^{204}\text{Pb}} = \frac{^{235}\text{U} \left(e^{\lambda_{235}t} - 1 \right)}{^{238}\text{U} \left(e^{\lambda_{238}t} - 1 \right)}$$

The isotope geology of lead



$$\left(\frac{^{206}\text{Pb}}{^{204}\text{Pb}} \right)_t = a_0 + \mu(e^{\lambda_1 T} - e^{\lambda_1 t})$$

$$\left(\frac{^{207}\text{Pb}}{^{204}\text{Pb}} \right)_t = b_0 + \frac{\mu}{137.88} (e^{\lambda_2 T} - e^{\lambda_2 t})$$

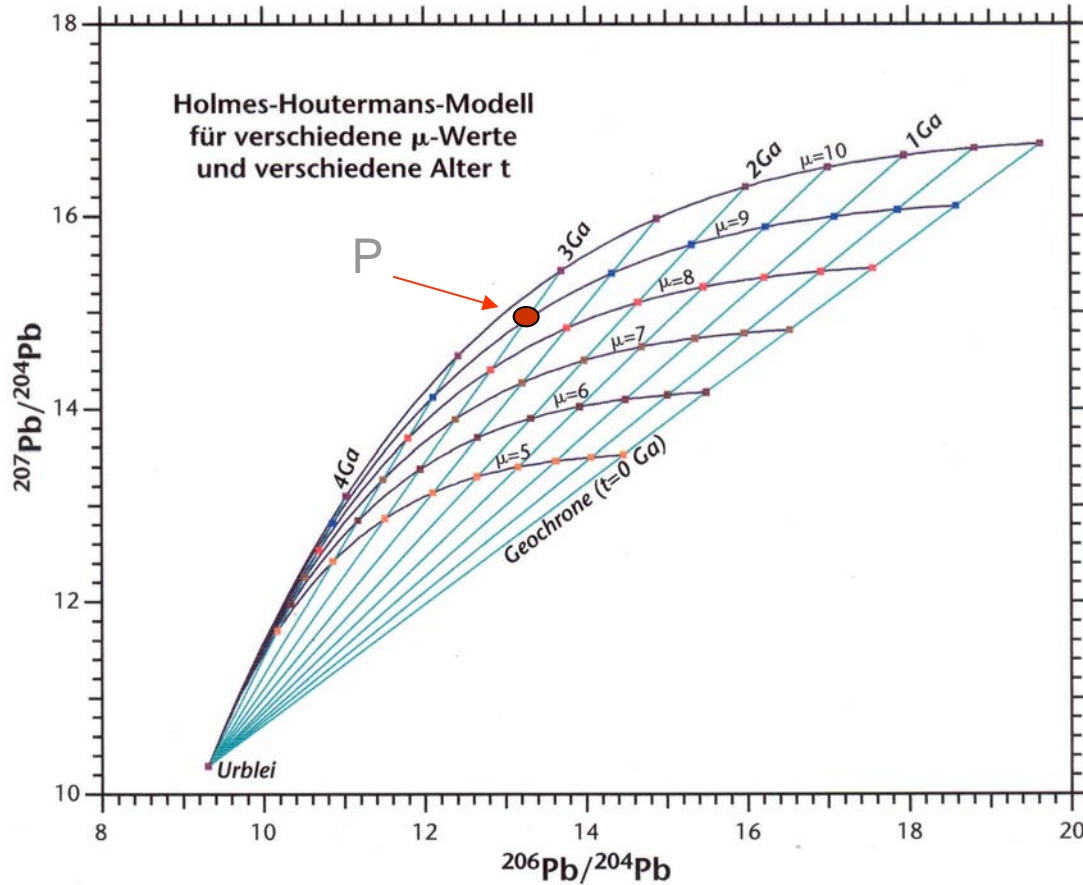
$$\left(\frac{^{206}\text{Pb}}{^{204}\text{Pb}} \right)_i = a_0 = 9.30$$

$$\left(\frac{^{207}\text{Pb}}{^{204}\text{Pb}} \right)_i = b_0 = 10.29$$

Dalrymple: "the hourglass of the solar system"

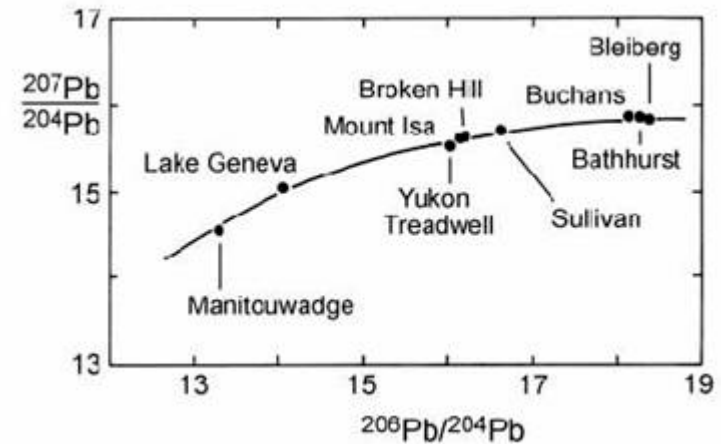
Primeval lead
(Isotope ratios of Pb in troilite of the iron meteorite Canyon Diablo)

The isotope geology of lead



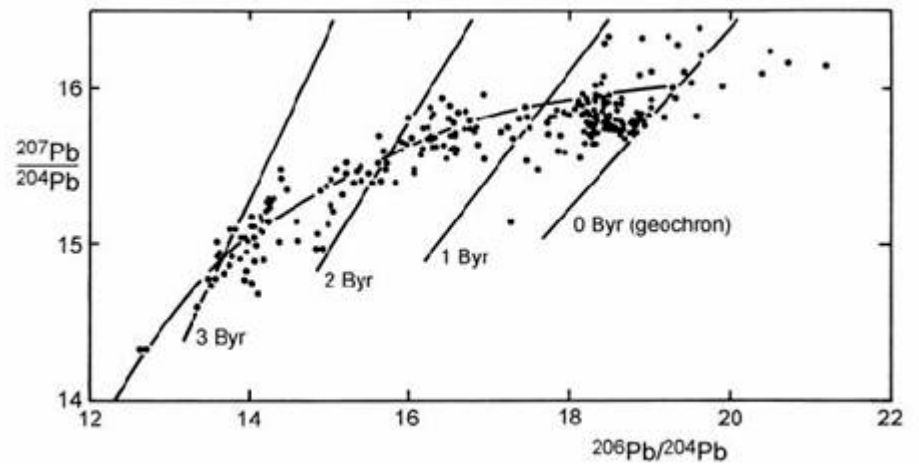
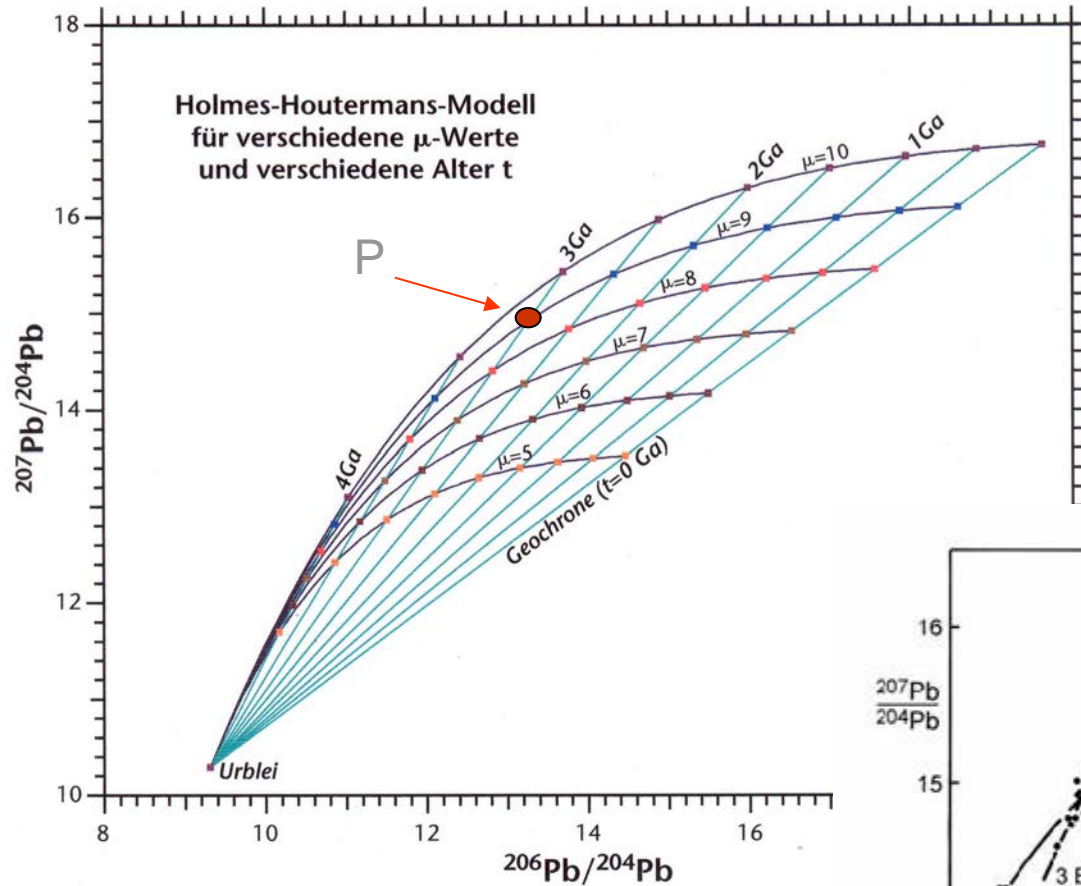
$$\left(\frac{^{206}\text{Pb}}{^{204}\text{Pb}} \right)_t = a_0 + \mu(e^{\lambda_1 T} - e^{\lambda_1 t})$$

$$\left(\frac{^{207}\text{Pb}}{^{204}\text{Pb}} \right)_t = b_0 + \frac{\mu}{137.88} (e^{\lambda_2 T} - e^{\lambda_2 t})$$

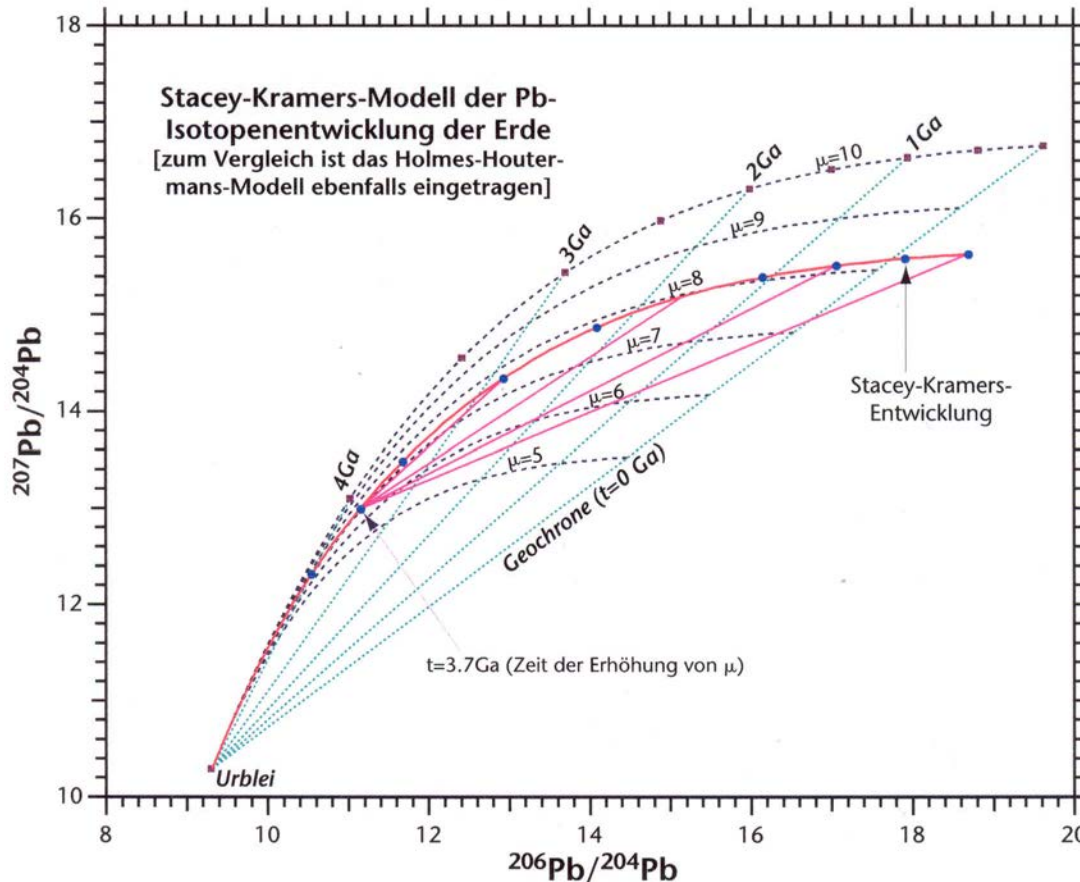


galena ores that form the basis of the 'conformable' Pb model

The isotope geology of lead



The isotope geology of Pb



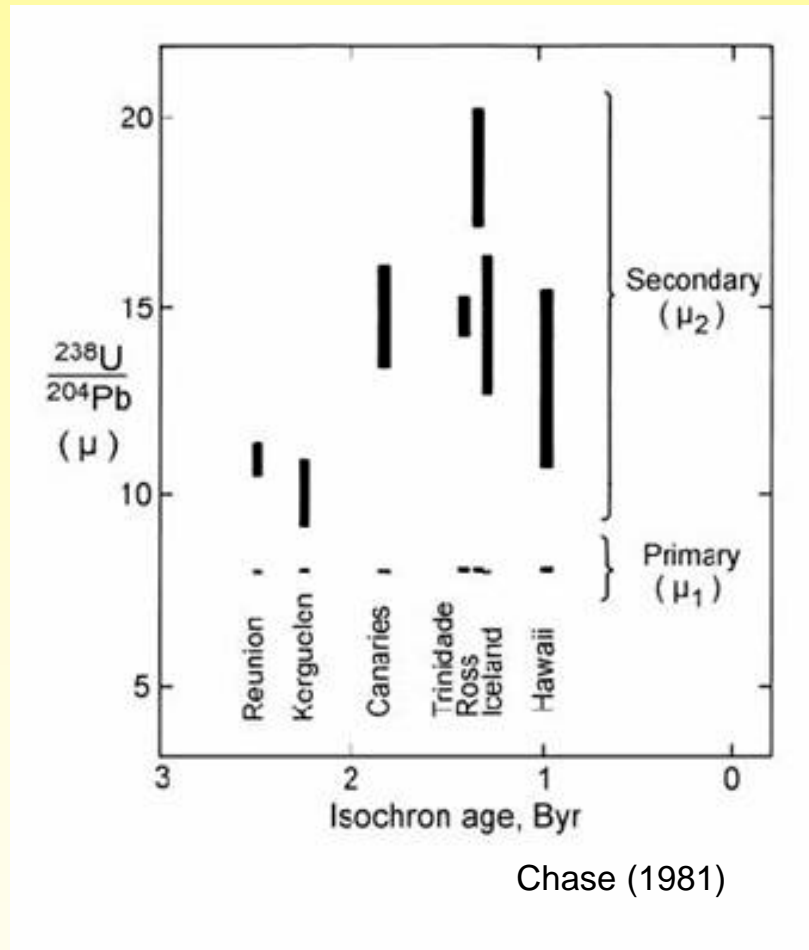
Two-stage Pb evolution (Stacey & Kramers 1975)

Pb evolves from primordial isotope ratios between 4.6 and 3.7 Ga in a reservoir with a μ -($^{238}\text{U}/^{204}\text{Pb}$) value of 7.2

At 3.7 Ga the μ -value of the reservoir was changed by geochemical differentiation to 9.7

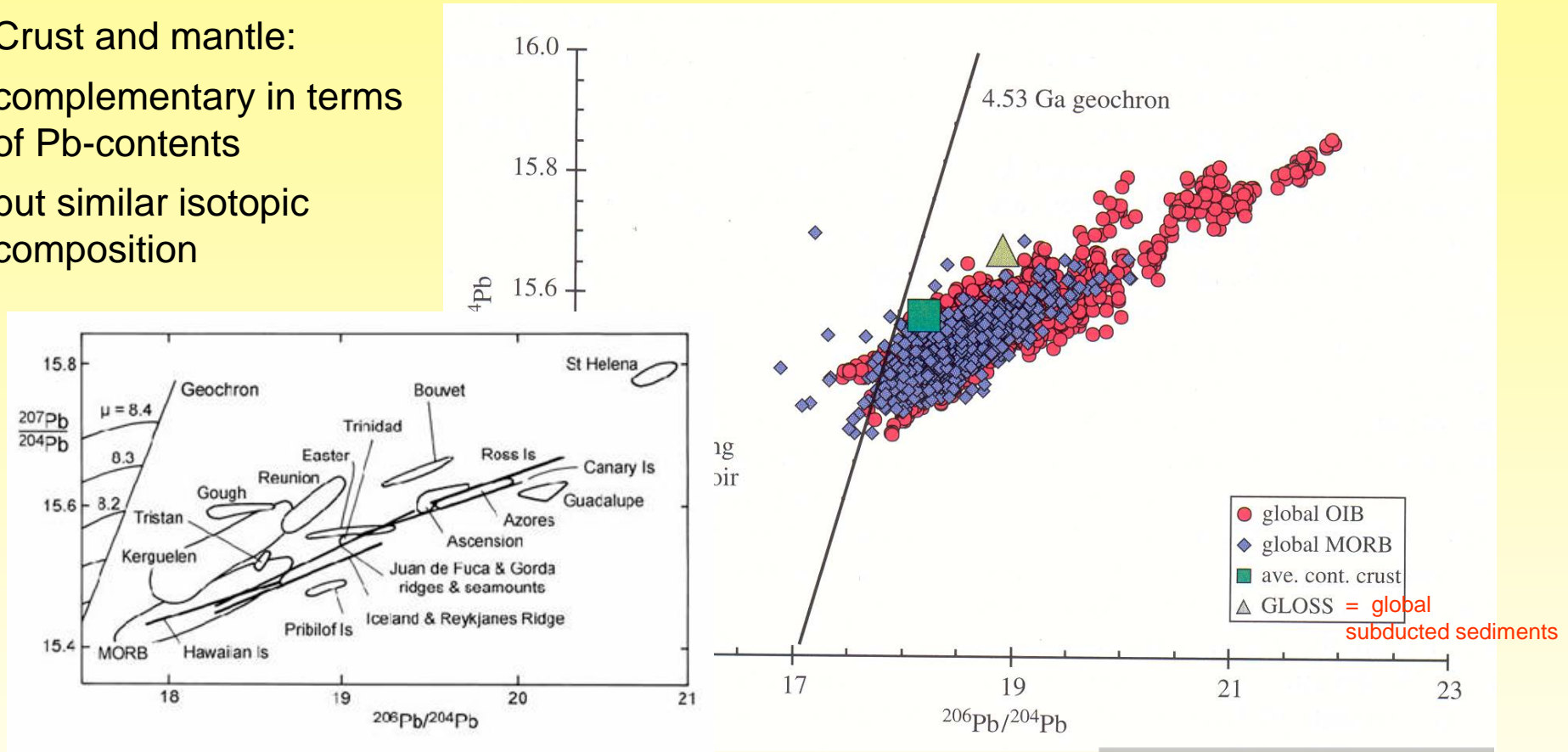
The isotope geology of Pb

Range of values required to explain OIB sources using a two-stage Pb evolution model. Parental mantle (μ_1) undergoes differentiation events at different times to yield discrete OIB source domains (μ_2).



The lead isotope paradox

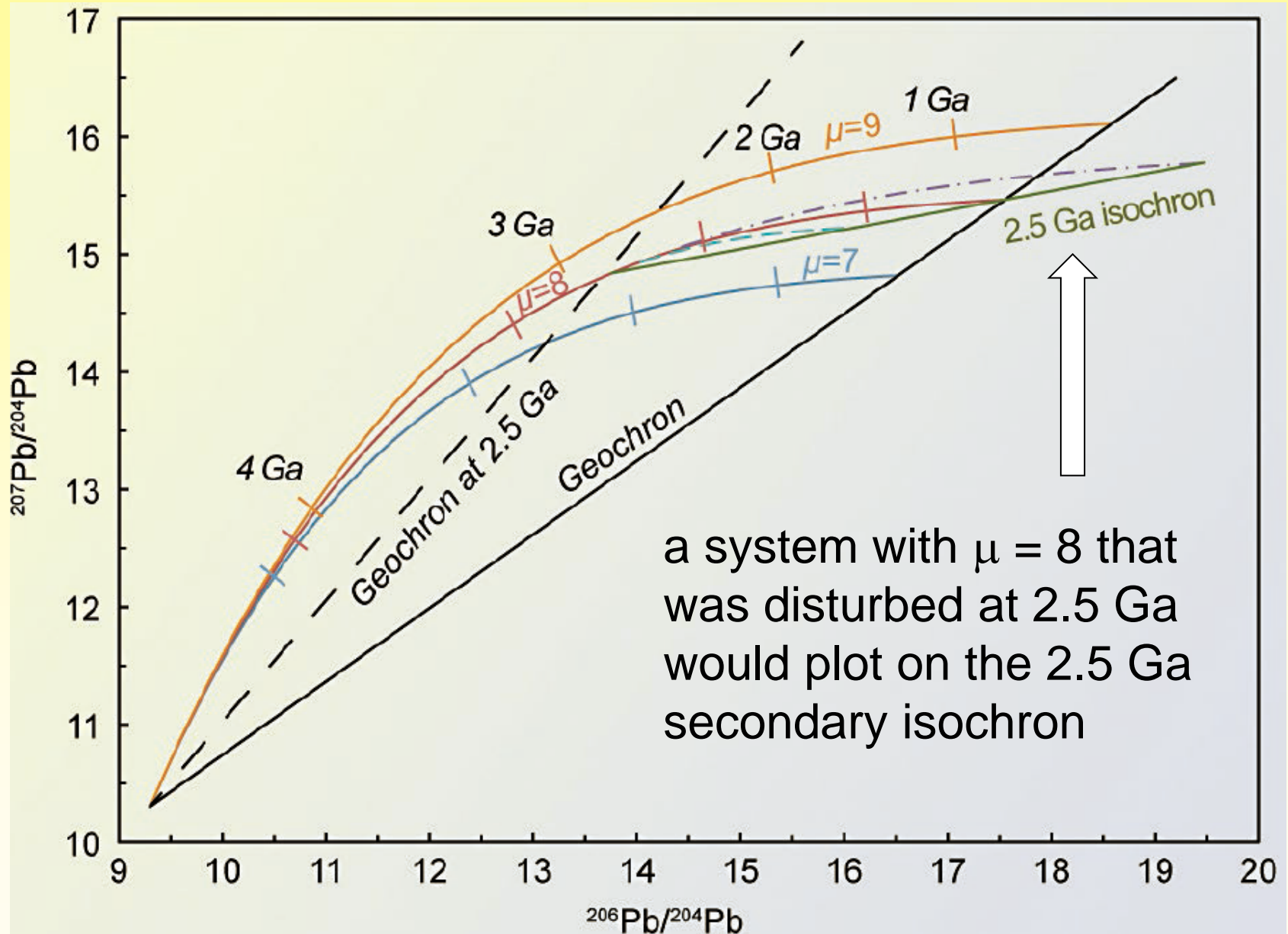
Crust and mantle:
complementary in terms
of Pb-contents
but similar isotopic
composition



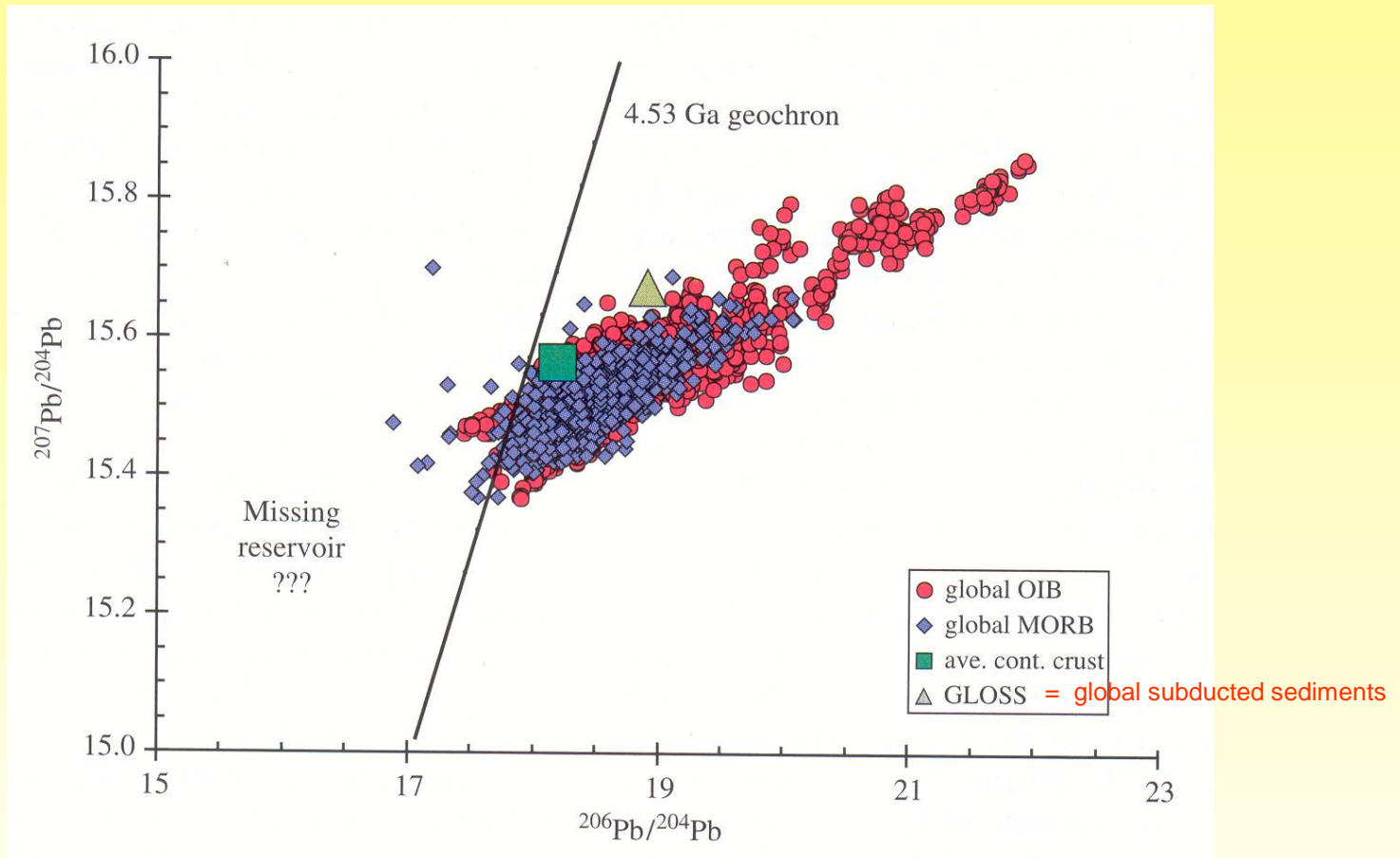
The fact that MORBs do not plot to the left of the geochron is called the “First terrestrial lead isotope paradox”

Hofmann (2003) :
Treatise on Geochemistry

The isotope geology of lead



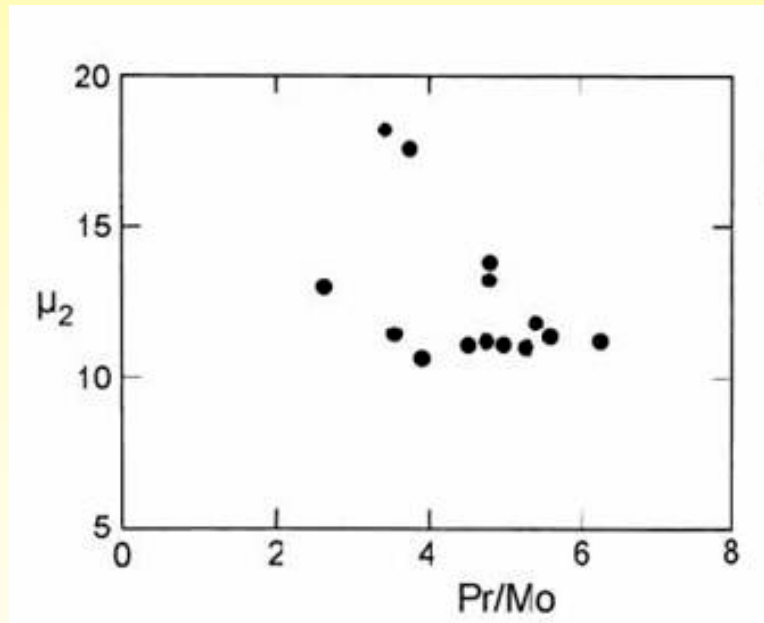
Das Pb-Paradox



- uptake of lead by the core (“core pumping”)

Das Pb-Paradox

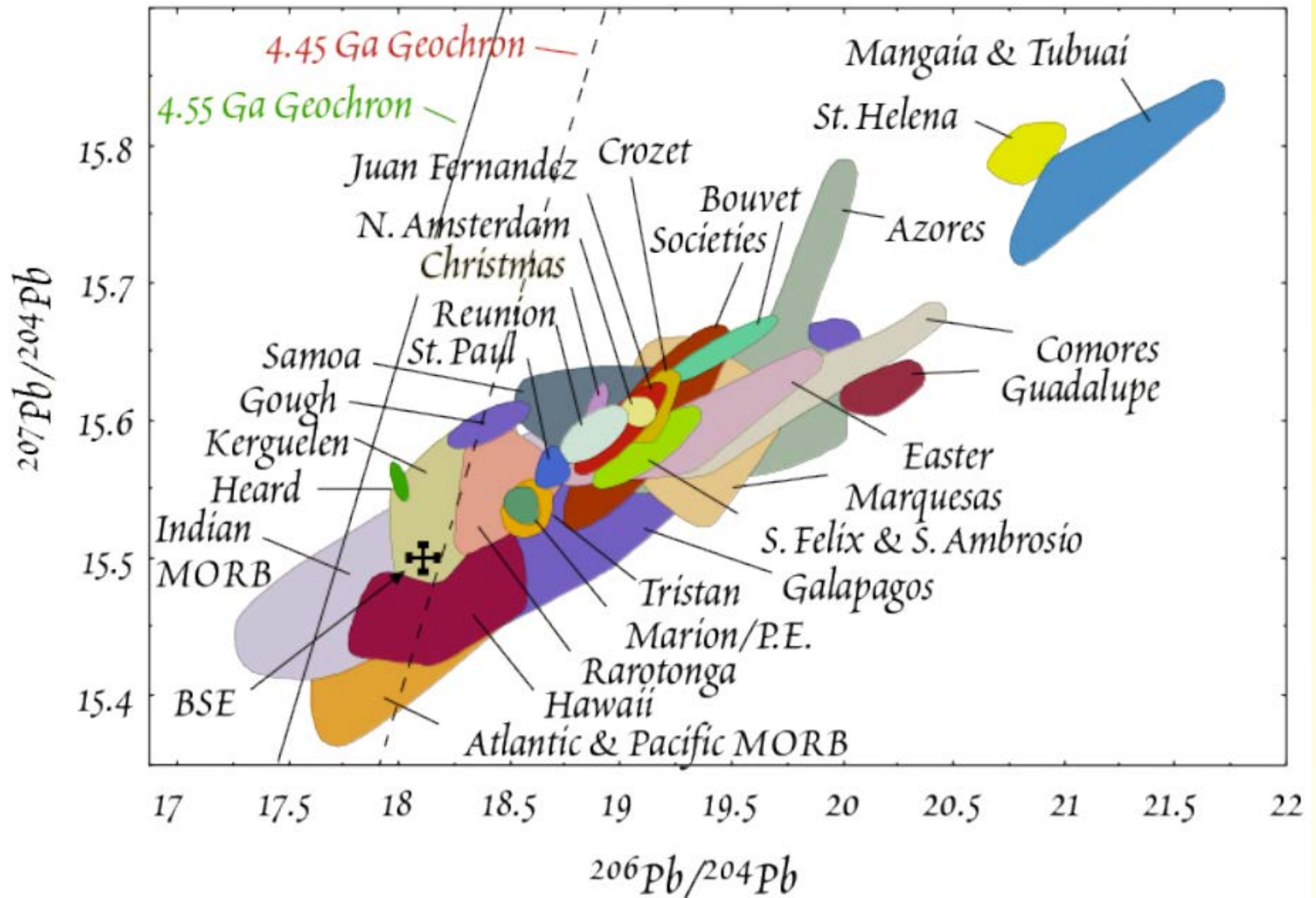
Core Pb-pumping model ([Allègre et al., 1982](#)) relies on the assumption that during core formation, Pb was partitioned into the core whereas U became enriched in the silicate Earth → increase of μ -value



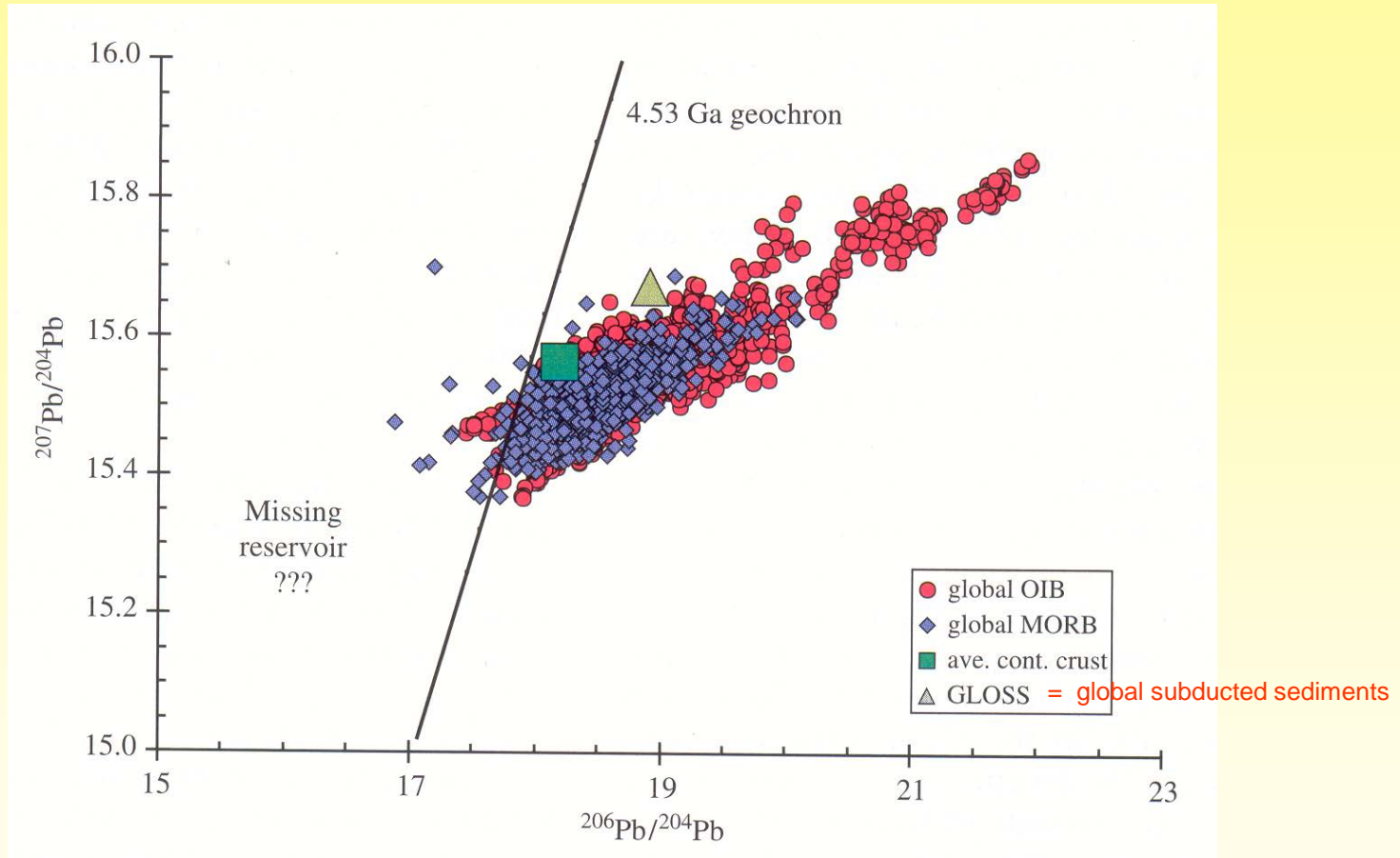
Lagos et al. (2008) The Earth's missing lead may not be in the core. Nature 456

Core pumping theory evaluated by Newsome et al. (1986)

Das Pb-Paradox

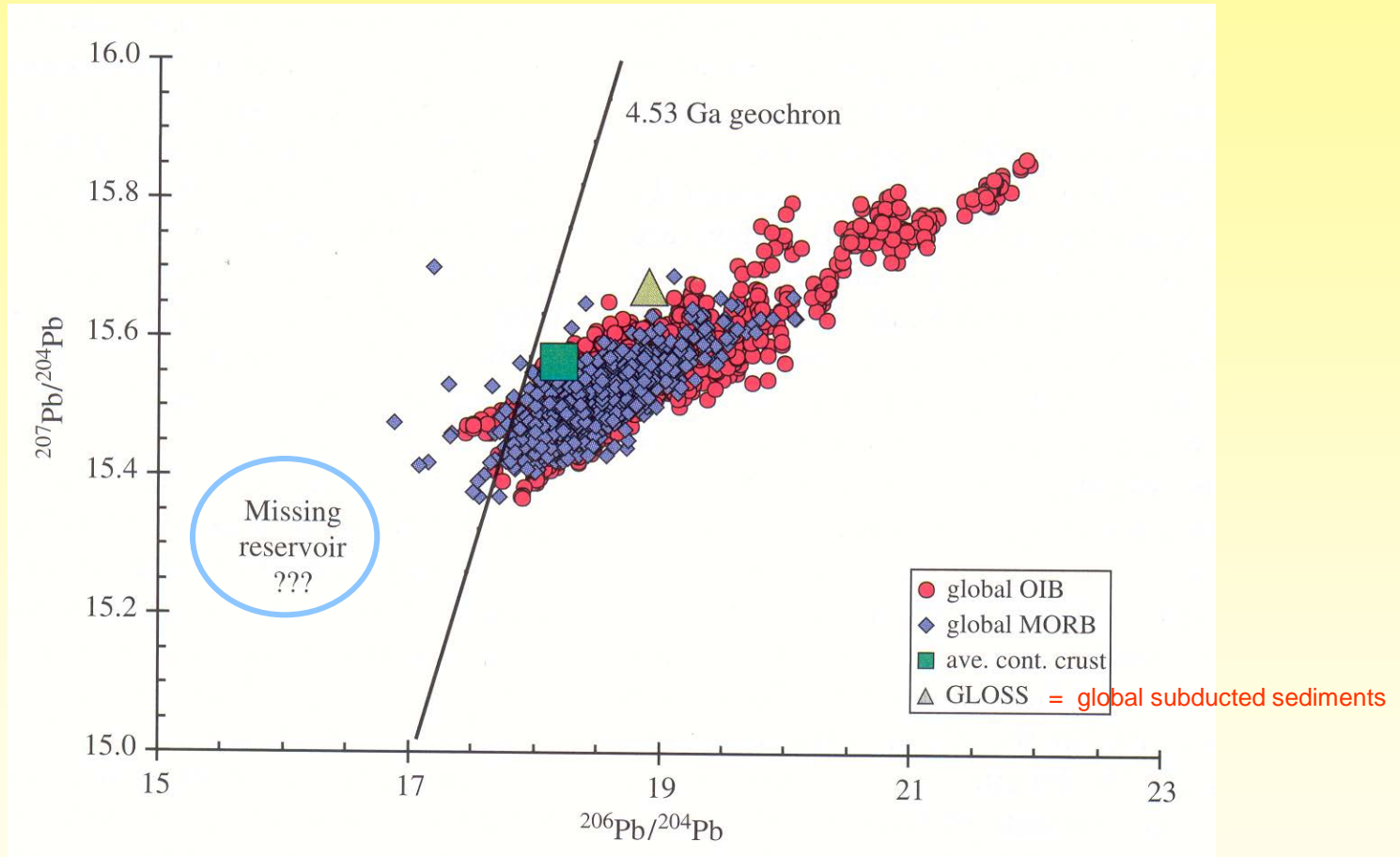


Das Pb-Paradox



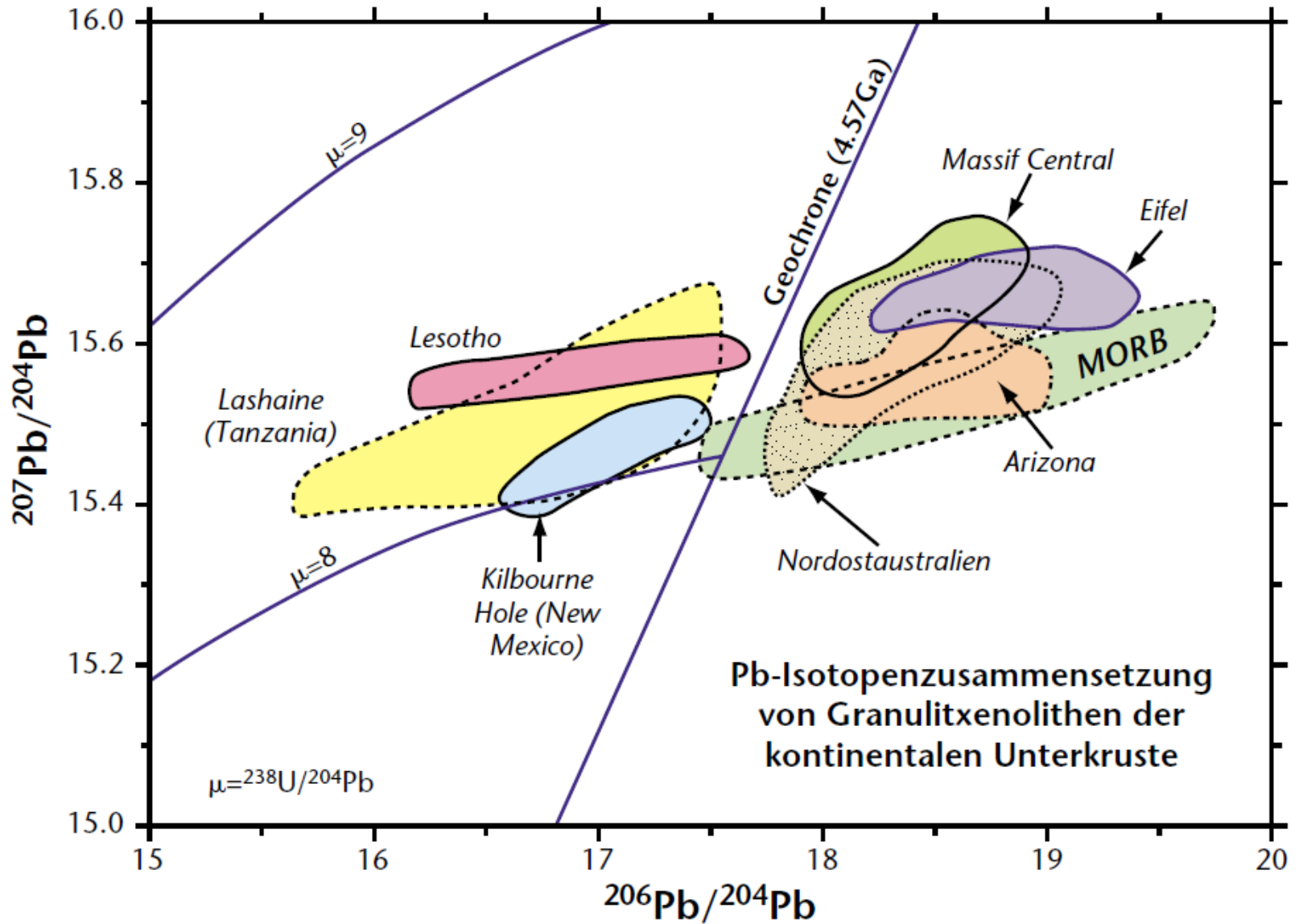
- storage of unradiogenic lead in the lower cont. crust or subcont. lithosphere

Das Pb-Paradox



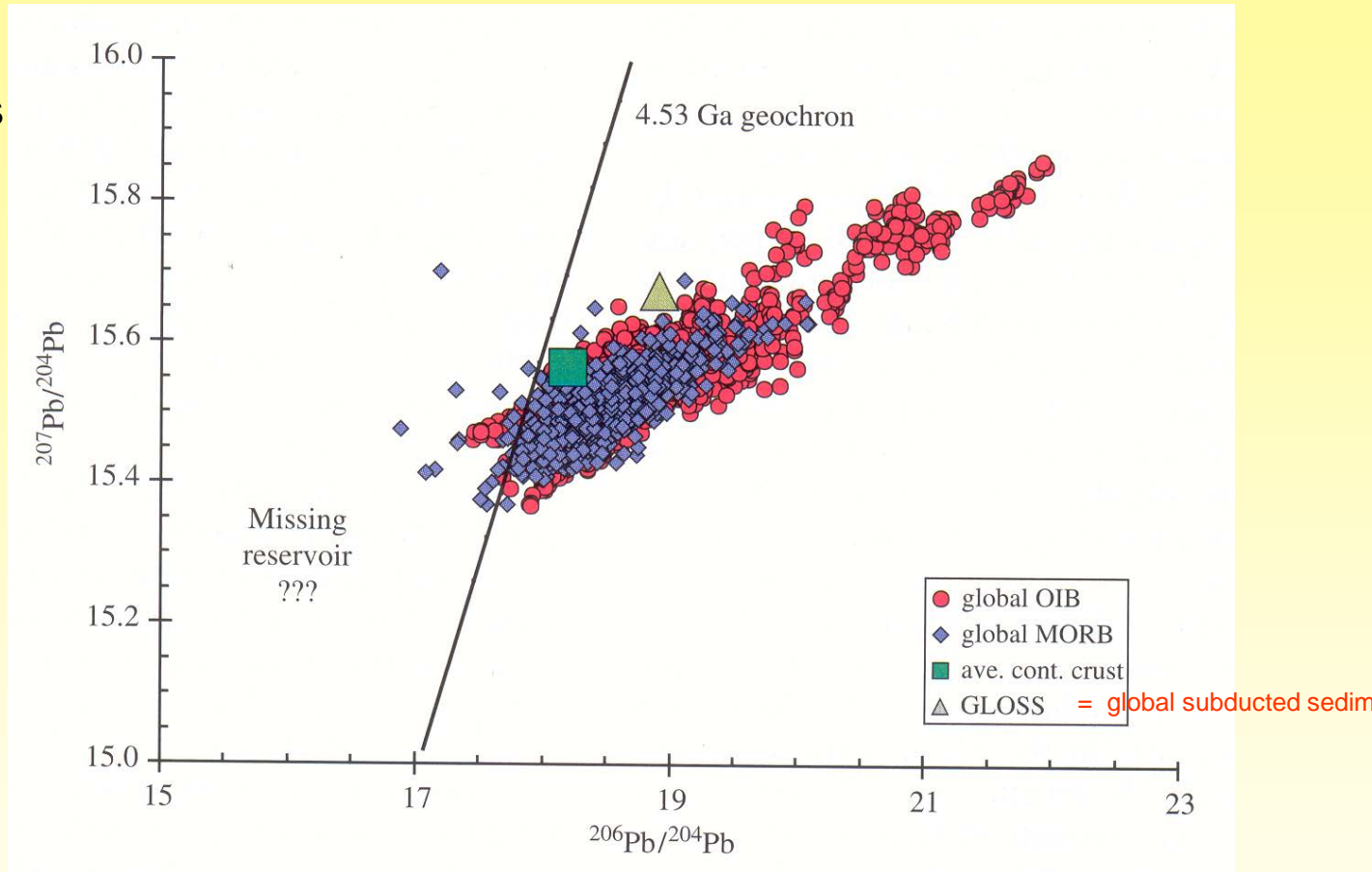
hidden reservoir with Pb isotopes to the left of the geochron

Das Pb Paradox



Das Pb-Paradox

Crust and mantle:
complementary in terms
of Pb-contents
but similar isotopic
composition



Ave. oceanic and cont. crust close to geochron
→ little net fractionation of U/Pb during
crust-mantle differentiation

The terrestrial Th/U ratio, κ

$$^{208}\text{Pb}^* = ^{232}\text{Th}(e^{\lambda_{232}t} - 1)$$

$$^{206}\text{Pb}^* = ^{238}\text{U}(e^{\lambda_{238}t} - 1)$$

$$\frac{^{208}\text{Pb}^*}{^{206}\text{Pb}^*} = \frac{^{232}\text{Th}}{^{238}\text{U}} = \kappa \frac{(e^{\lambda_{232}t} - 1)}{(e^{\lambda_{238}t} - 1)}$$

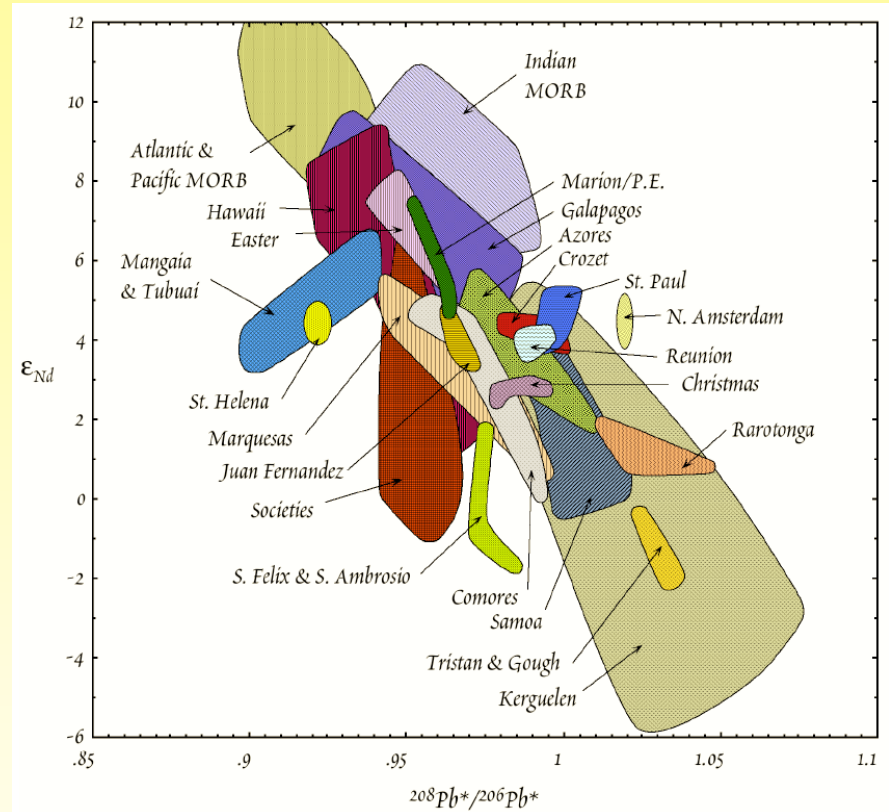
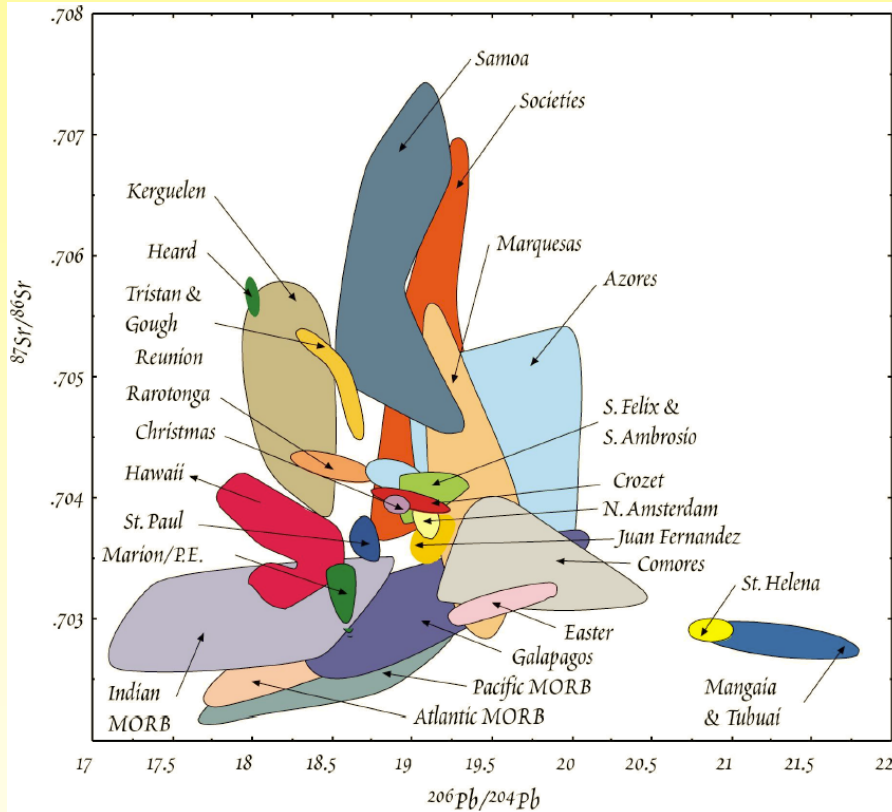
$$\frac{^{208}\text{Pb}^*/^{206}\text{Pb}^*}{\kappa} = \frac{(^{208}\text{Pb}/^{204}\text{Pb})_t - (^{208}\text{Pb}/^{204}\text{Pb})_T}{(^{206}\text{Pb}/^{204}\text{Pb})_t - (^{206}\text{Pb}/^{204}\text{Pb})_T}$$

T = initial age/age of the Earth

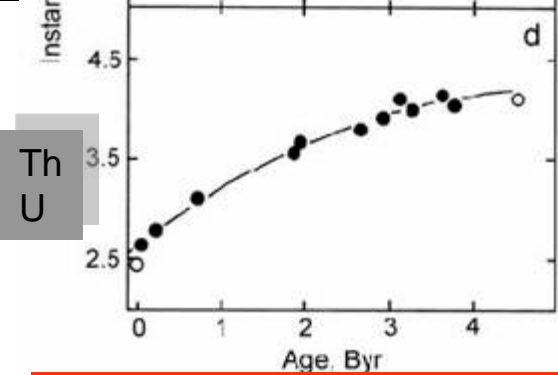
t = formation age of sample

* = radiogenic (from ^{238}U and ^{232}Th) Pb

Pb-Isotopie und Th/U-Verhältnis, κ



Die fehlende Korrelation zwischen Pb und Sr ist auf das anomale Verhalten von Blei zurückzuführen!

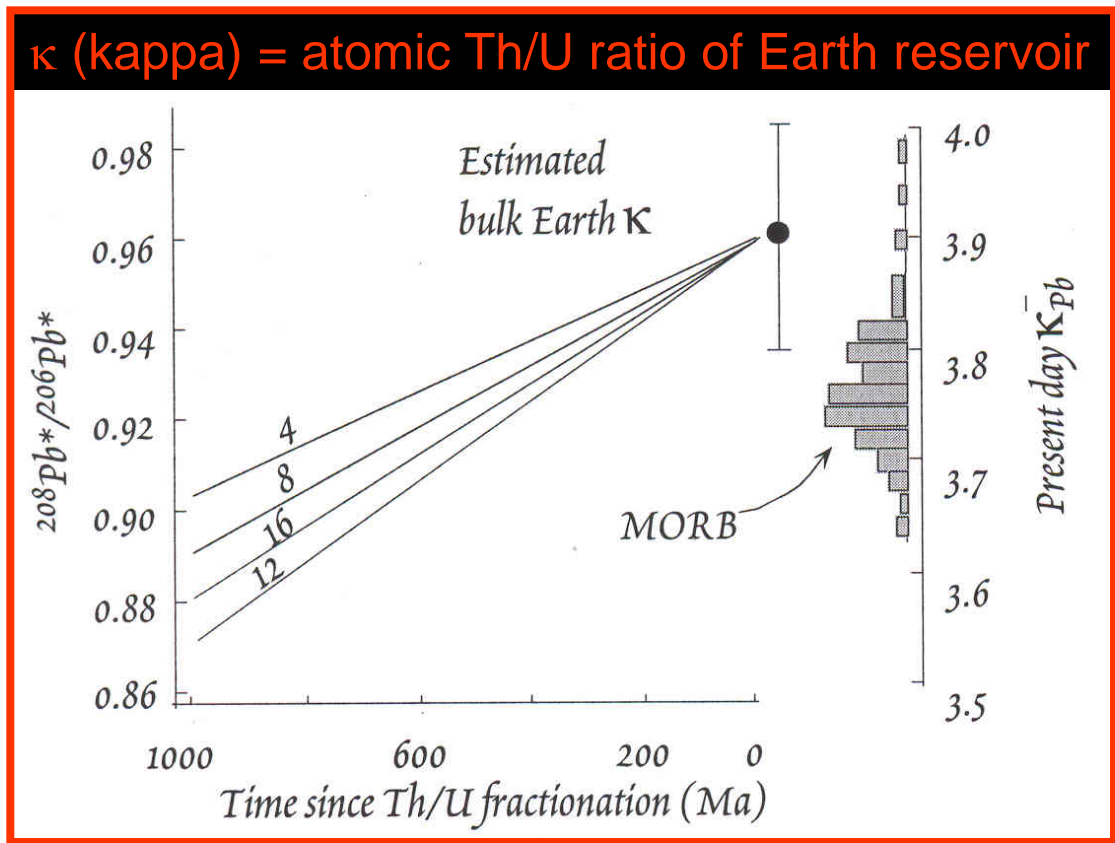


Open system model of Pb isotope evolution of the Earth

Time integrated Th/U ratio (derived from Pb isotope data) of ~3.75 in MORB is much higher than the „instantaneous“ present-day Th/U ratio of ~2.5!!

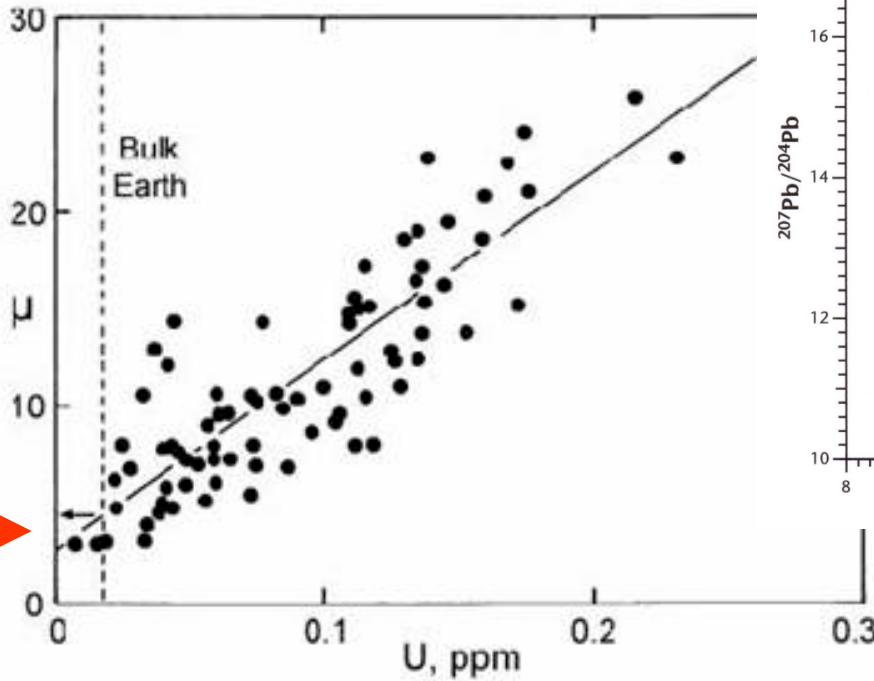
→ MORB reservoir is buffered over geological time by a less depleted reservoir, i.e:

→ MORB source had a brief residence time in the depleted reservoir and spend most of Earth history in a reservoir with a Th/U ratio near Bulk Earth.



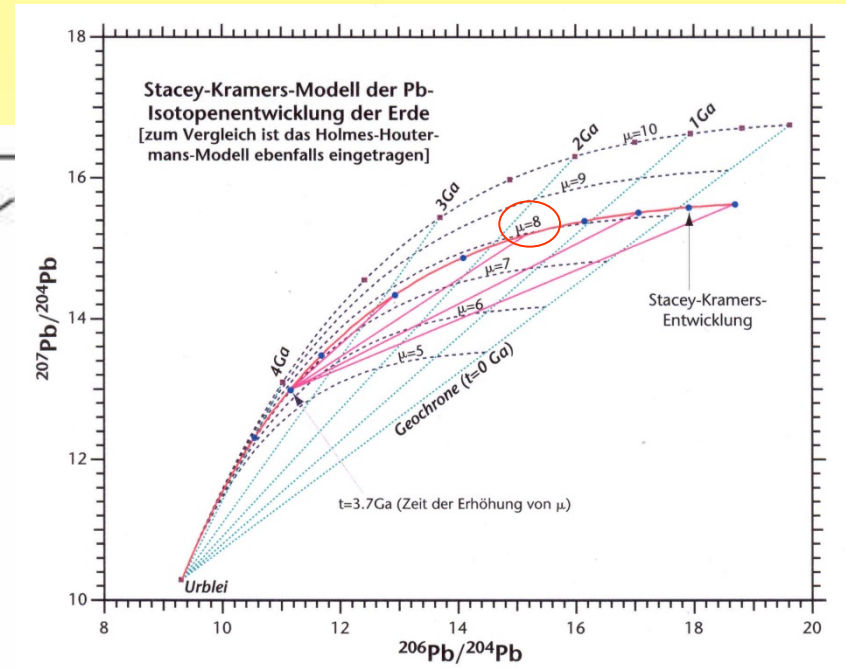
The upper mantle μ -value

$$\frac{^{238}\text{U}}{^{204}\text{Pb}}$$



$\mu = 4.5$

Instantaneous μ value in MORB source



Übung:

Das Alter der Erde

In der Stratigraphie eingesetzte geochemische Methoden

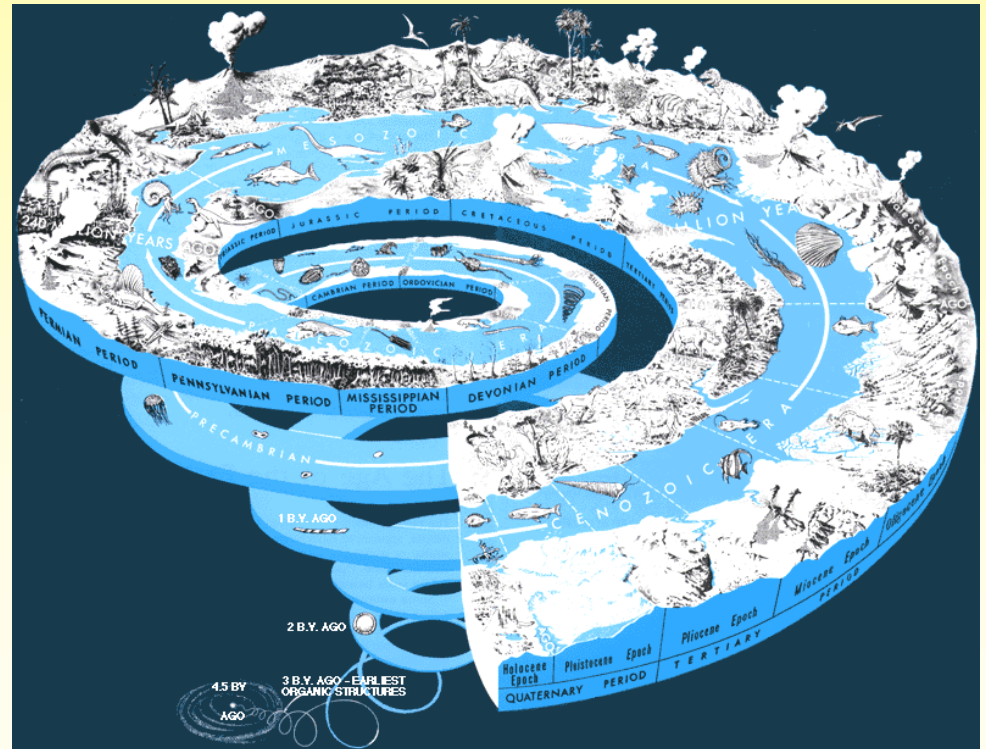
Stratigraphie + Datierung = Chronostratigraphie

alle anderen Methoden der
Stratigraphie werden in die
chronostratigraphische Abfolge
eingehängt.

How do we know the age of the Earth?

Radiometric
dating

A time machine to
the past



Elements

An International Magazine of Mineralogy, Geochemistry, and Petrology

February 2013
Volume 9, Number 1

ISSN 1811-5209

One Hundred Years of Geochronology

DANIEL J. CONDON and MARK D. SCHMITZ, Guest Editors

...and Counting

Precision and Accuracy in Geochronology

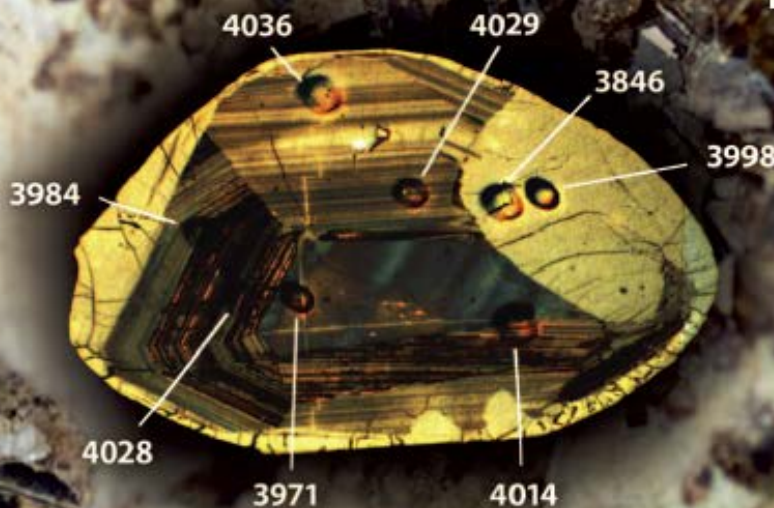
High-Precision Geochronology

High-Spatial-Resolution Geochronology

**Dating the Oldest Rocks
in the Solar System**

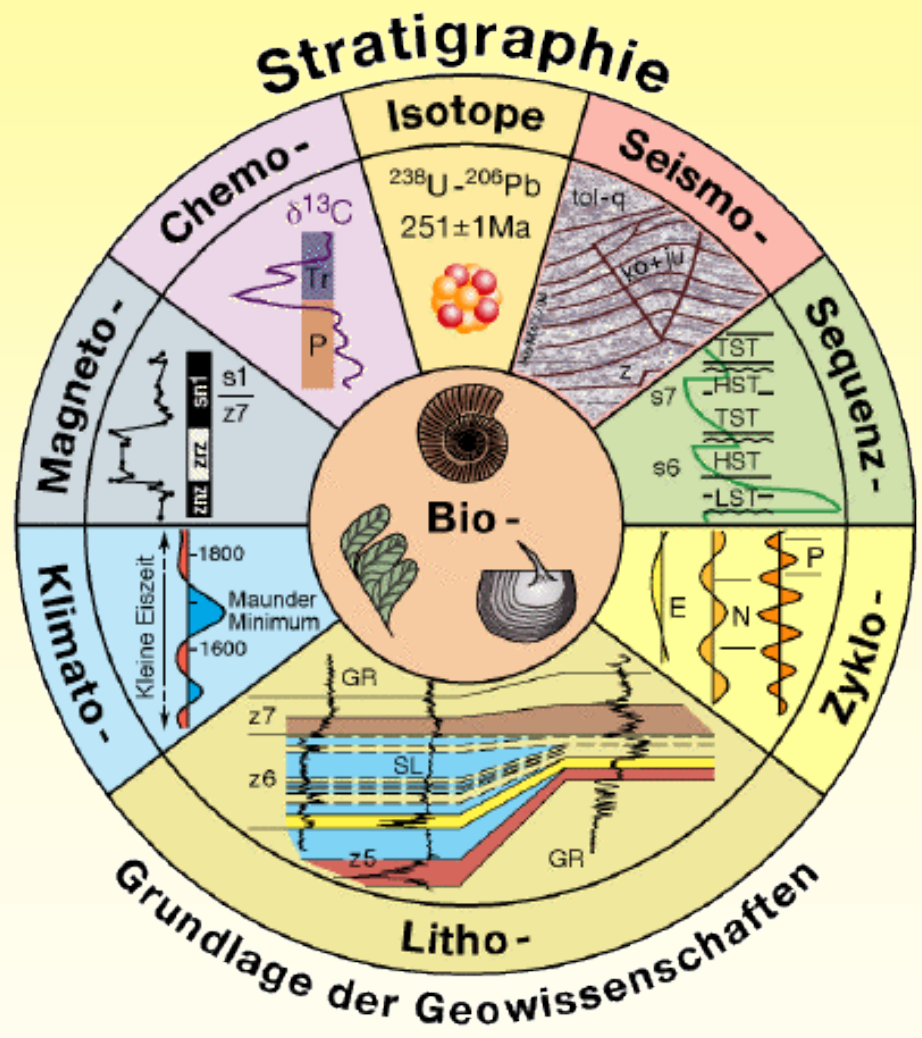
**Time Constraints in the
Quaternary Period**

**100 Years of U-Pb
Geochronology**



Arthur Holmes 1913:
The age of the Earth

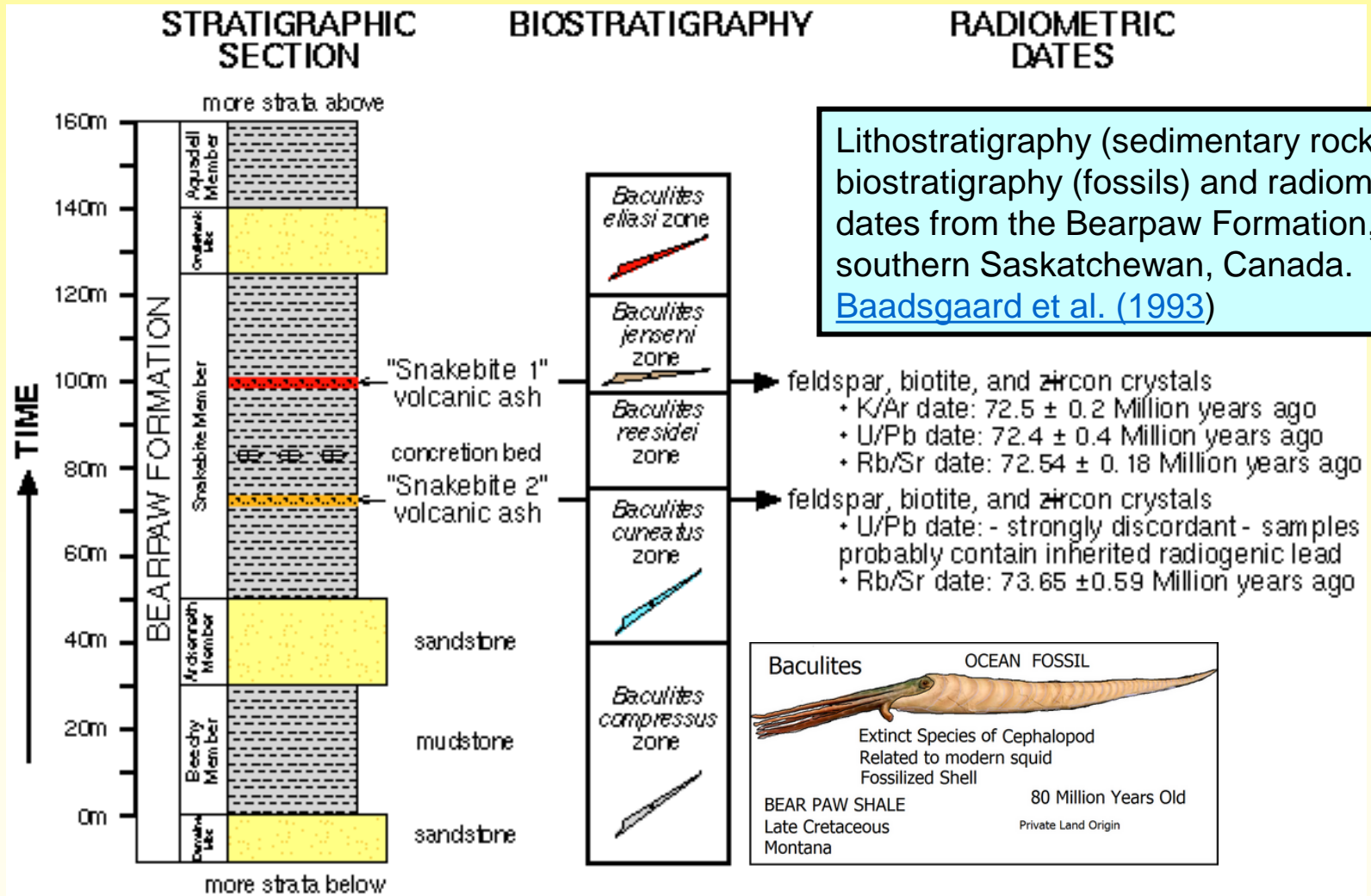
Stratigraphie und Datierung (Chronostratigraphie)





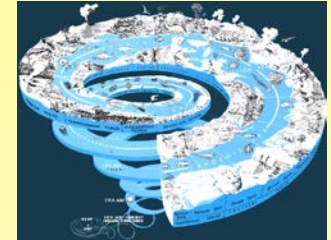
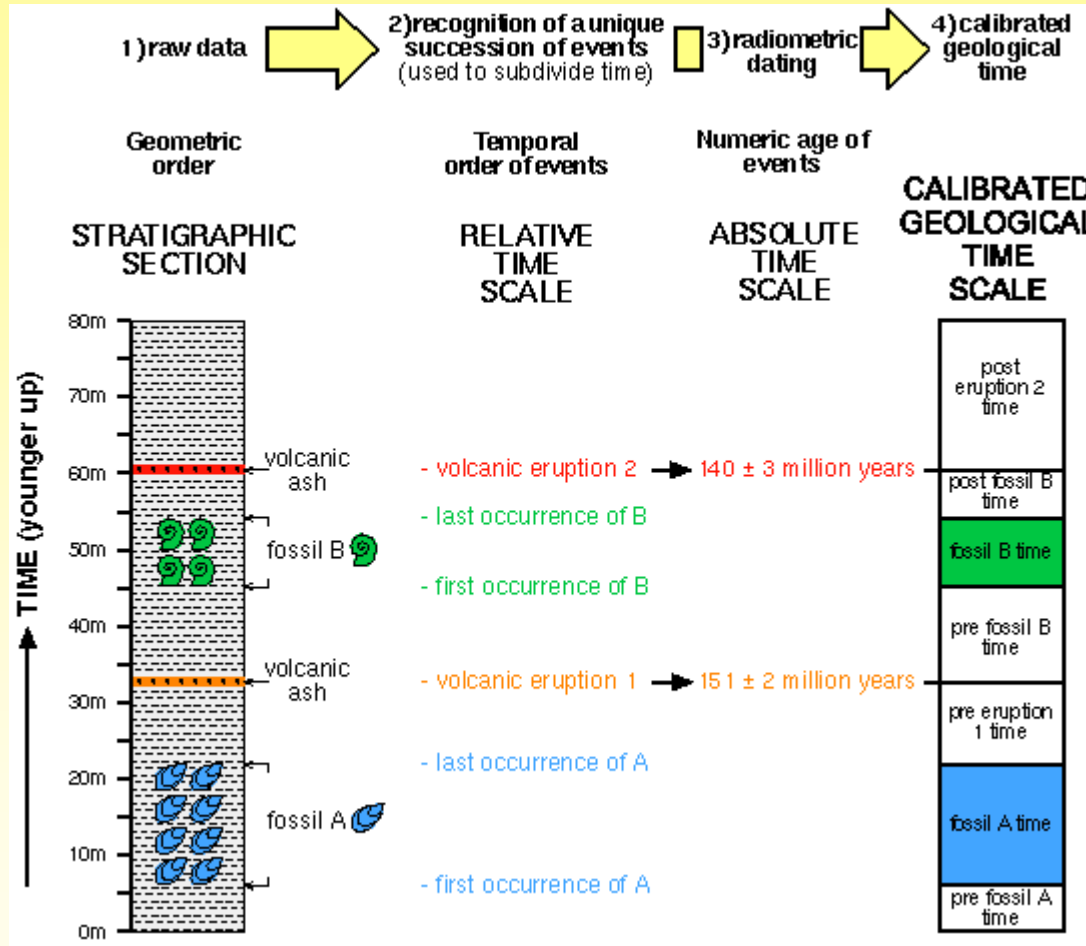
Jutulhogget, Antarctica (photo H. Frimmel)

Towards a calibrated geological time scale



Lithostratigraphy (sedimentary rocks), biostratigraphy (fossils) and radiometric dates from the Bearpaw Formation, southern Saskatchewan, Canada. [Baadsgaard et al. \(1993\)](#)

Towards a calibrated geological time scale



Radiogene Isotopensysteme

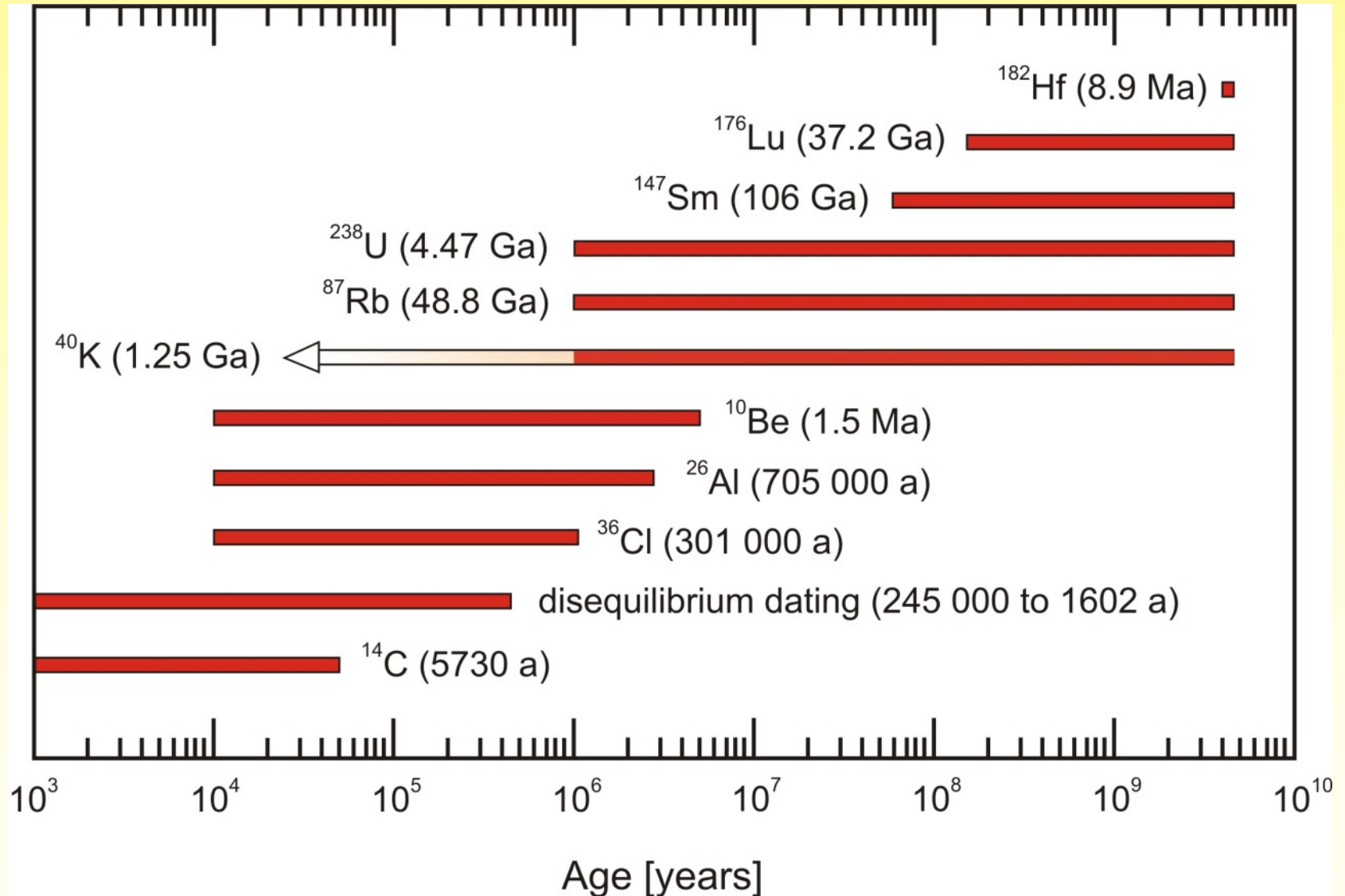
Goldschmidt-Klassifikation im Periodensystem der Elemente

	1																	18	
1	1 H	2												5 B	6 C	7 N	8 O	9 F	10 Ne
2	3 Li	4 Be												13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
3	11 Na	12 Mg	3	4	5	6	7	8	9	10	11	12	13 Al	14 Si	15 P	16 S	17 Cl	18 Ar	
4	19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr	
5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	(43) Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe	
6	55 Cs	56 Ba	57-71 Lan	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn	
7	87 Fr	88 Ra	89-103 Act	(104) Rf	(105) Db	(106) Sg	(107) Bh	(108) Hs	(109) Mt	(110) Ds	(111) Rg	(112) Cn	(113) Uut	(114) Uuq	(115) Uup	(116) Uuh	(117) Uus	(118) Uuo	
Lanthanoide	57 La	58 Ce	59 Pr	60 Nd	(61) Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu				
Actinoide	89 Ac	90 Th	91 Pa	92 U	(93) Np	(94) Pu	(95) Am	(96) Cm	(97) Bk	(98) Cf	(99) Es	(100) Fm	(101) Md	(102) No	(103) Lr				

Legende:

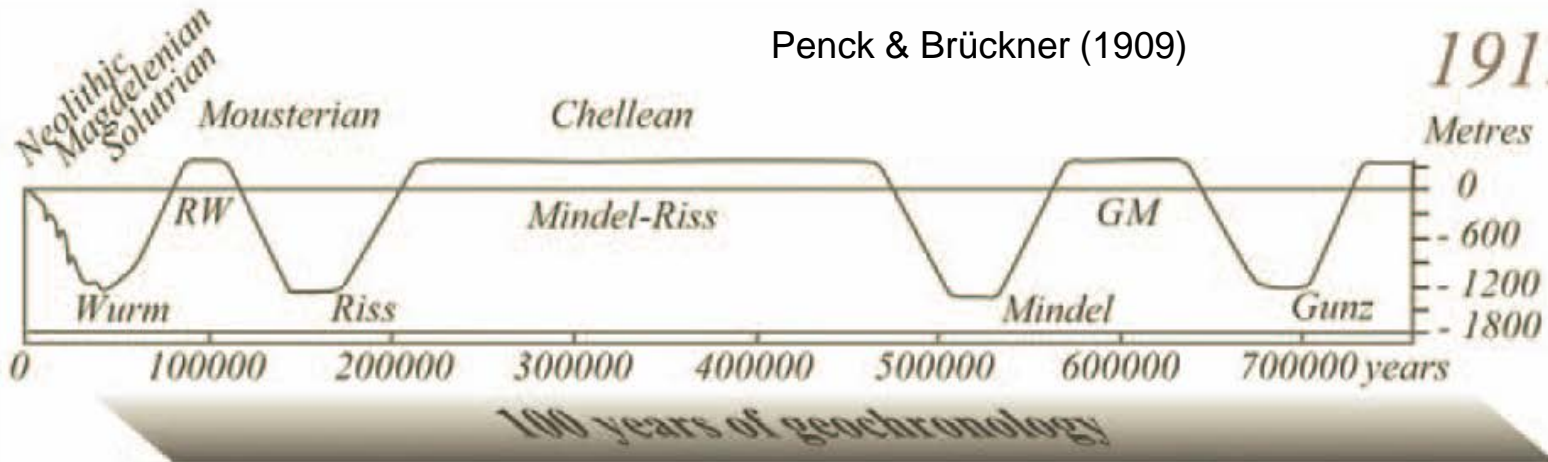
Atmosphärophil	Chalcophil	Lithophil	Siderophil	sehr selten
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Datierungssysteme

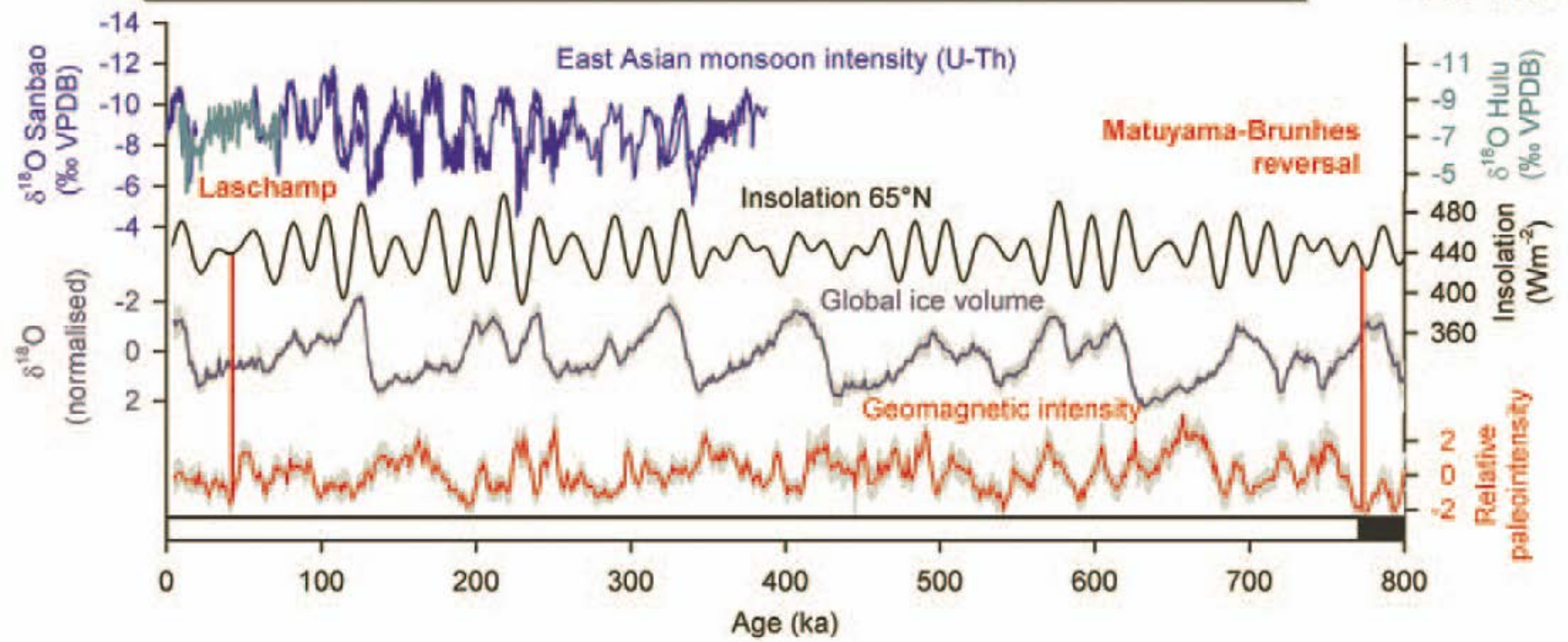


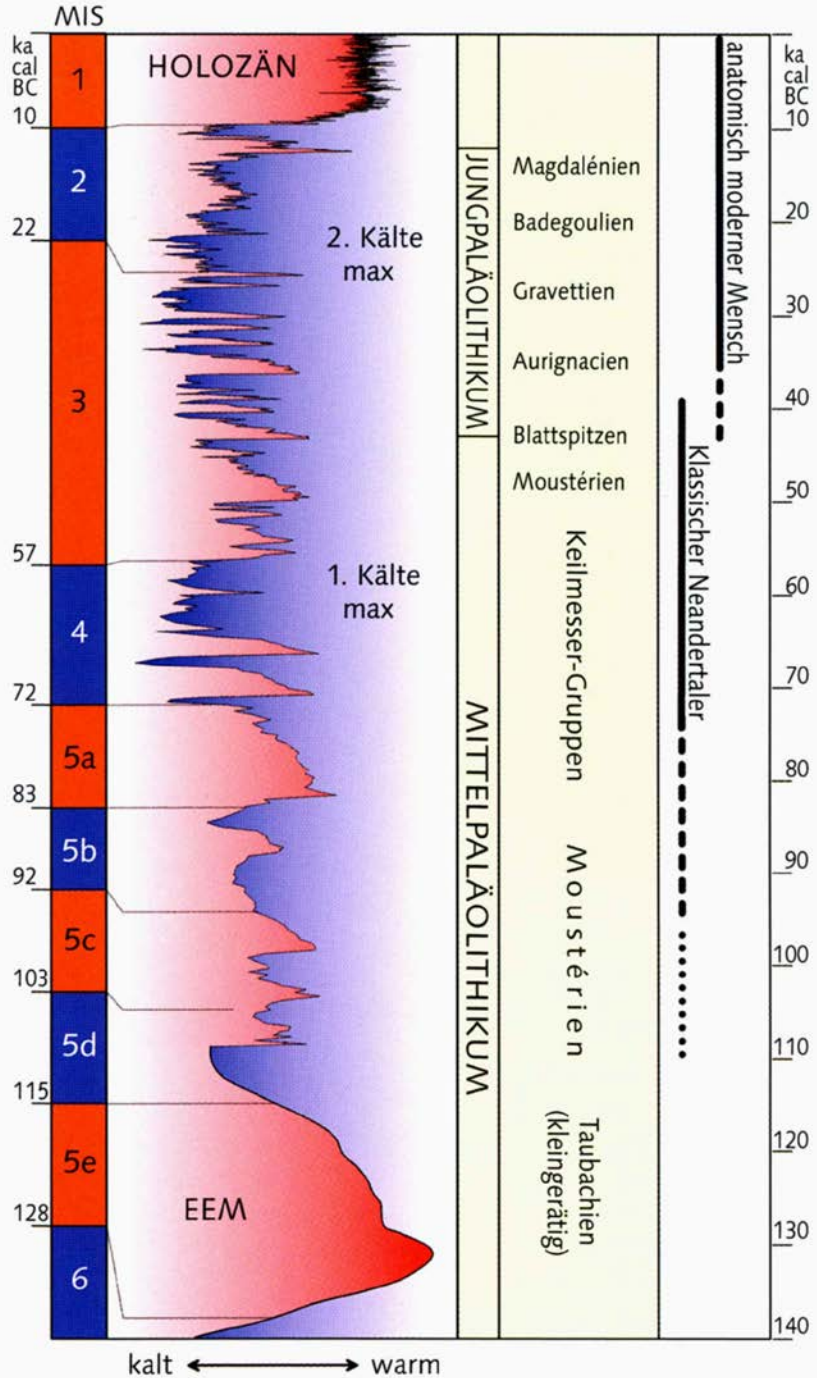
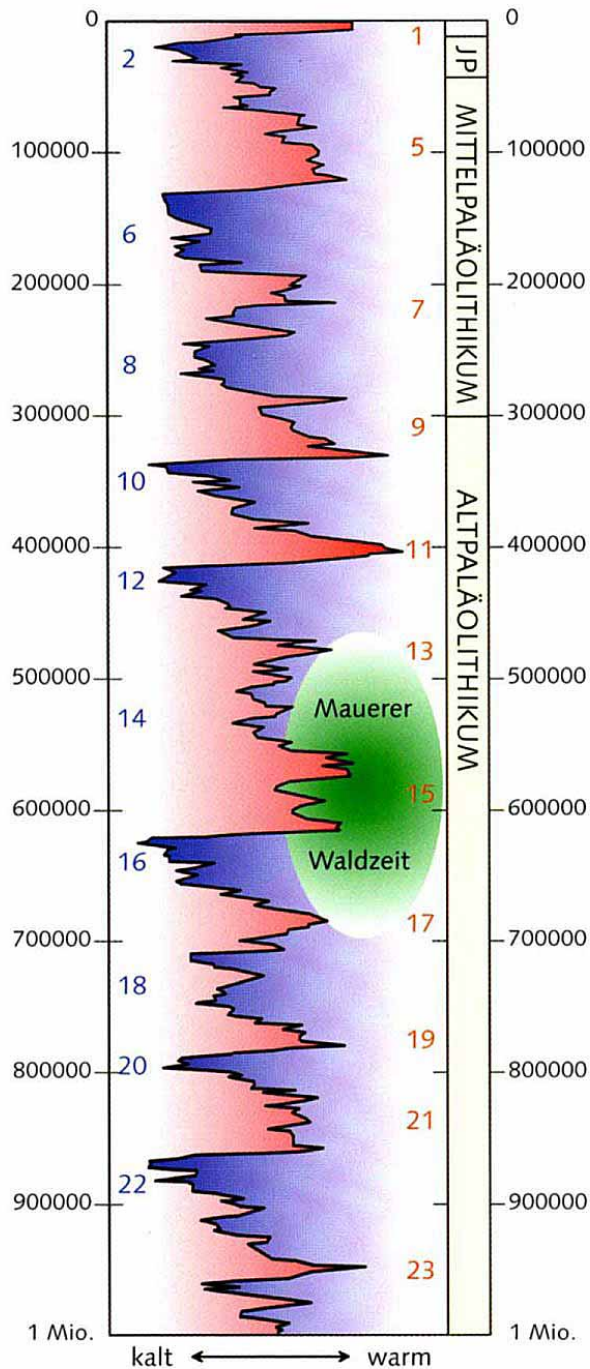
Penck & Brückner (1909)

1913



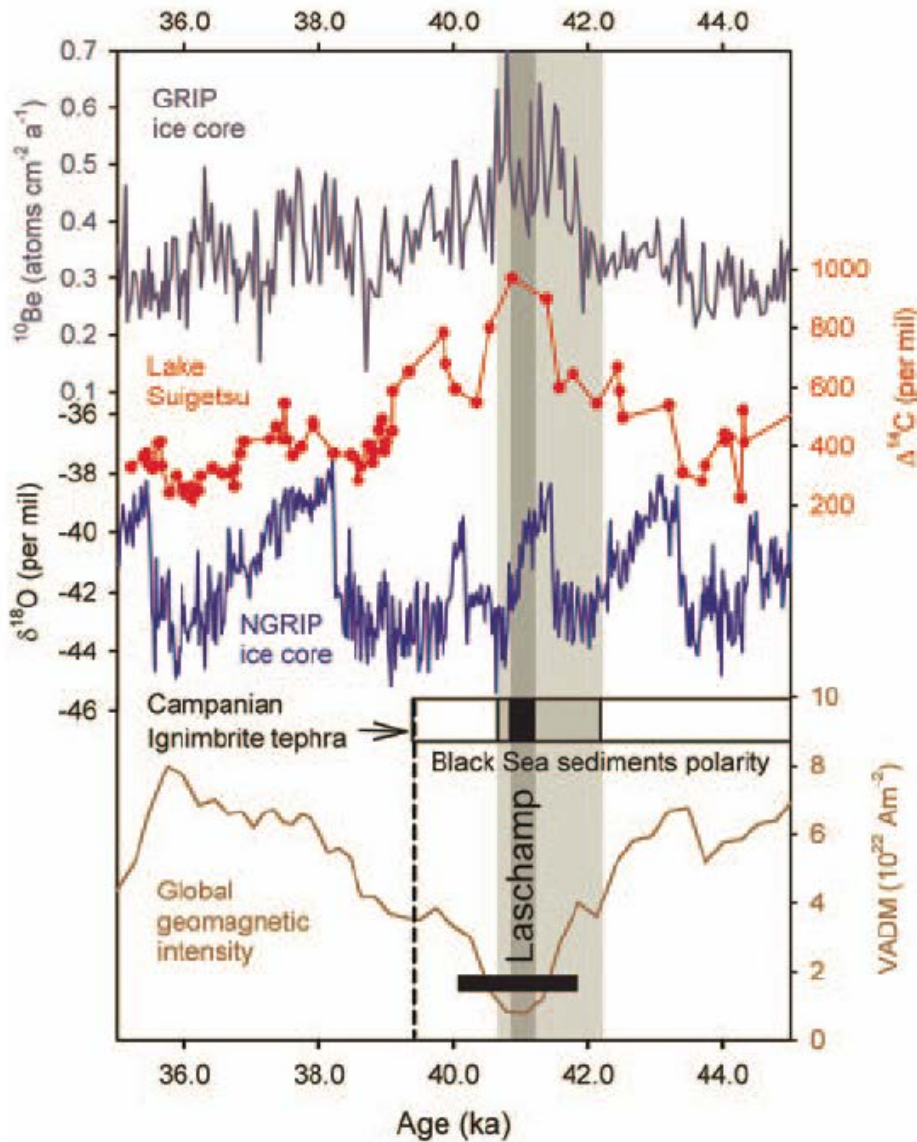
2013





Laschamp

the Earth at 40 ka

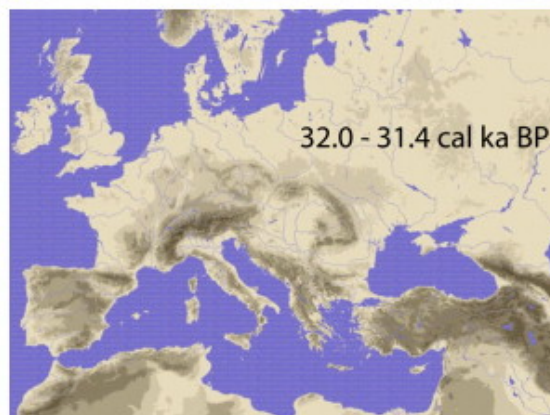
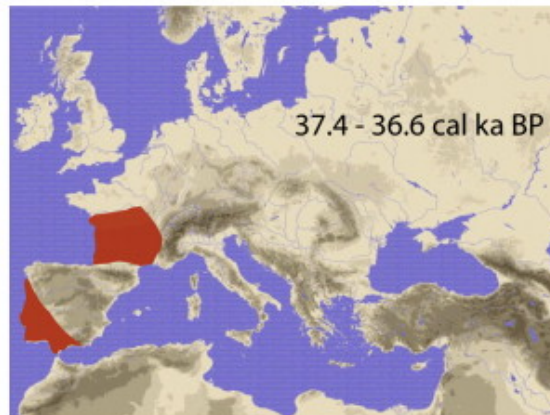
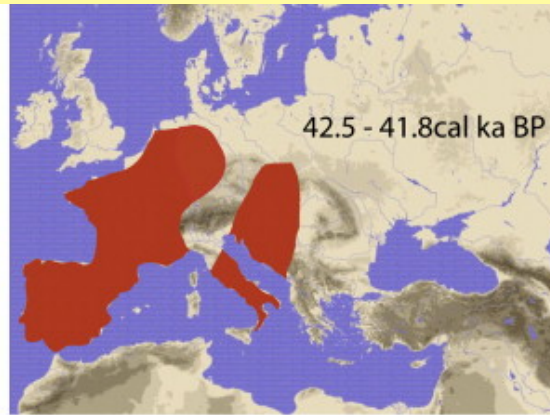
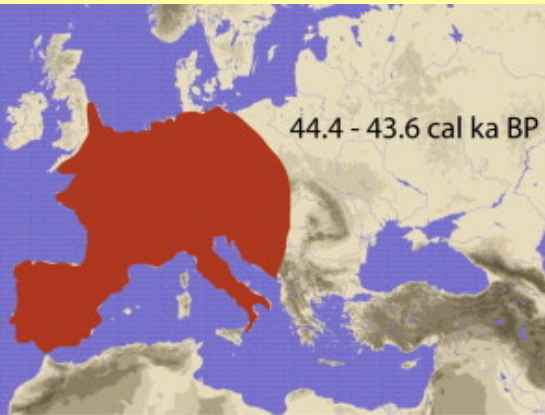


short and fast reversal of the Earth's magnetic field

short-term climate variability of the last ice age

and volcanic eruption in Italy

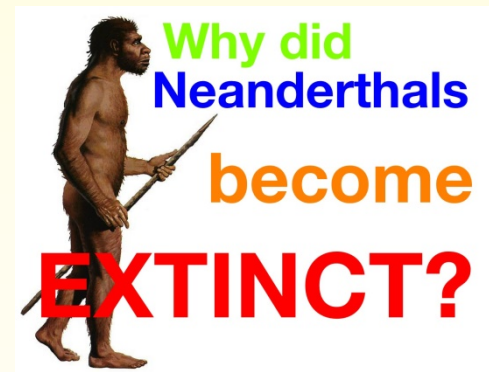
Das Laschamp Ereignis



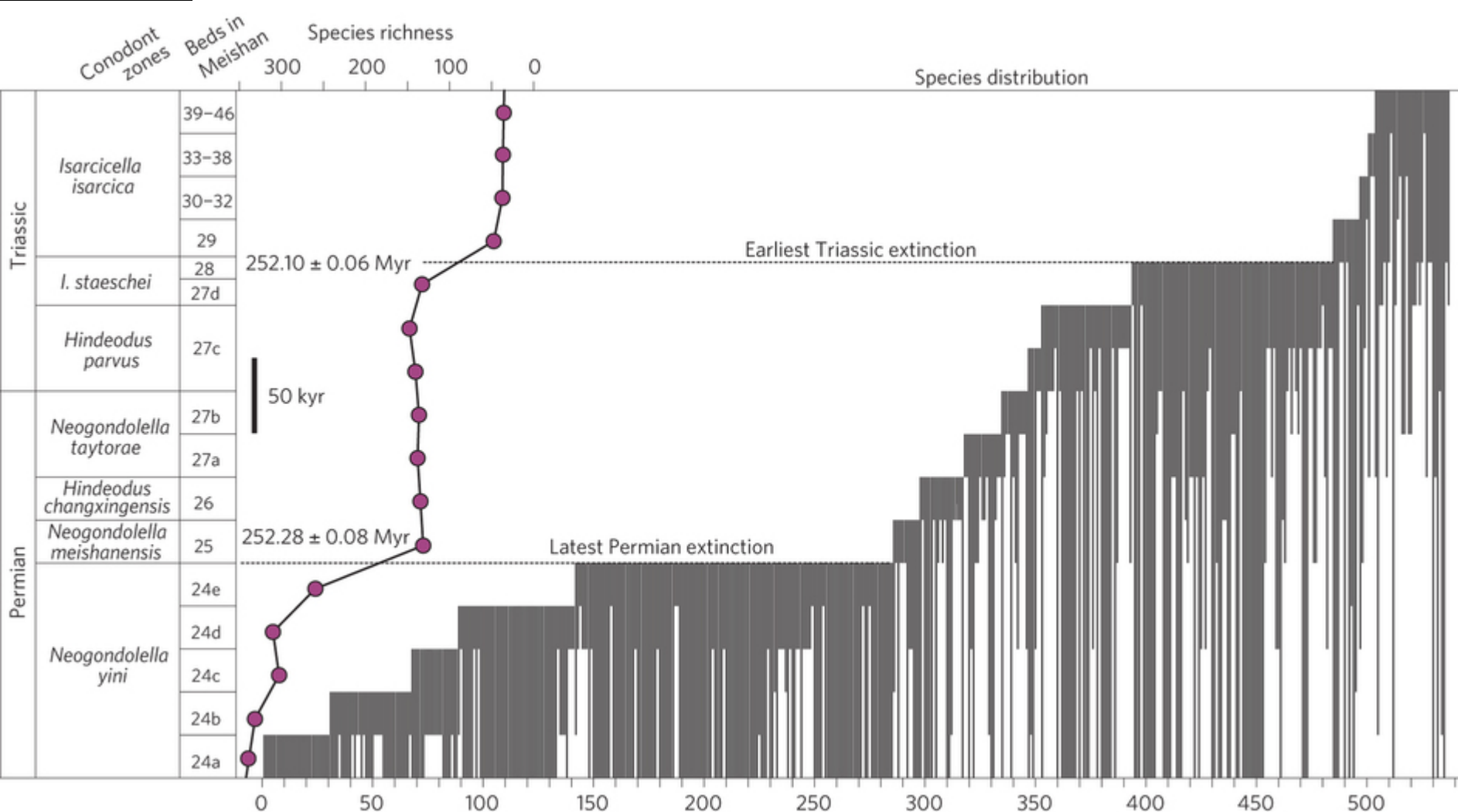
short and fast reversal of the Earth's magnetic field

short-term climate variability of the last ice age

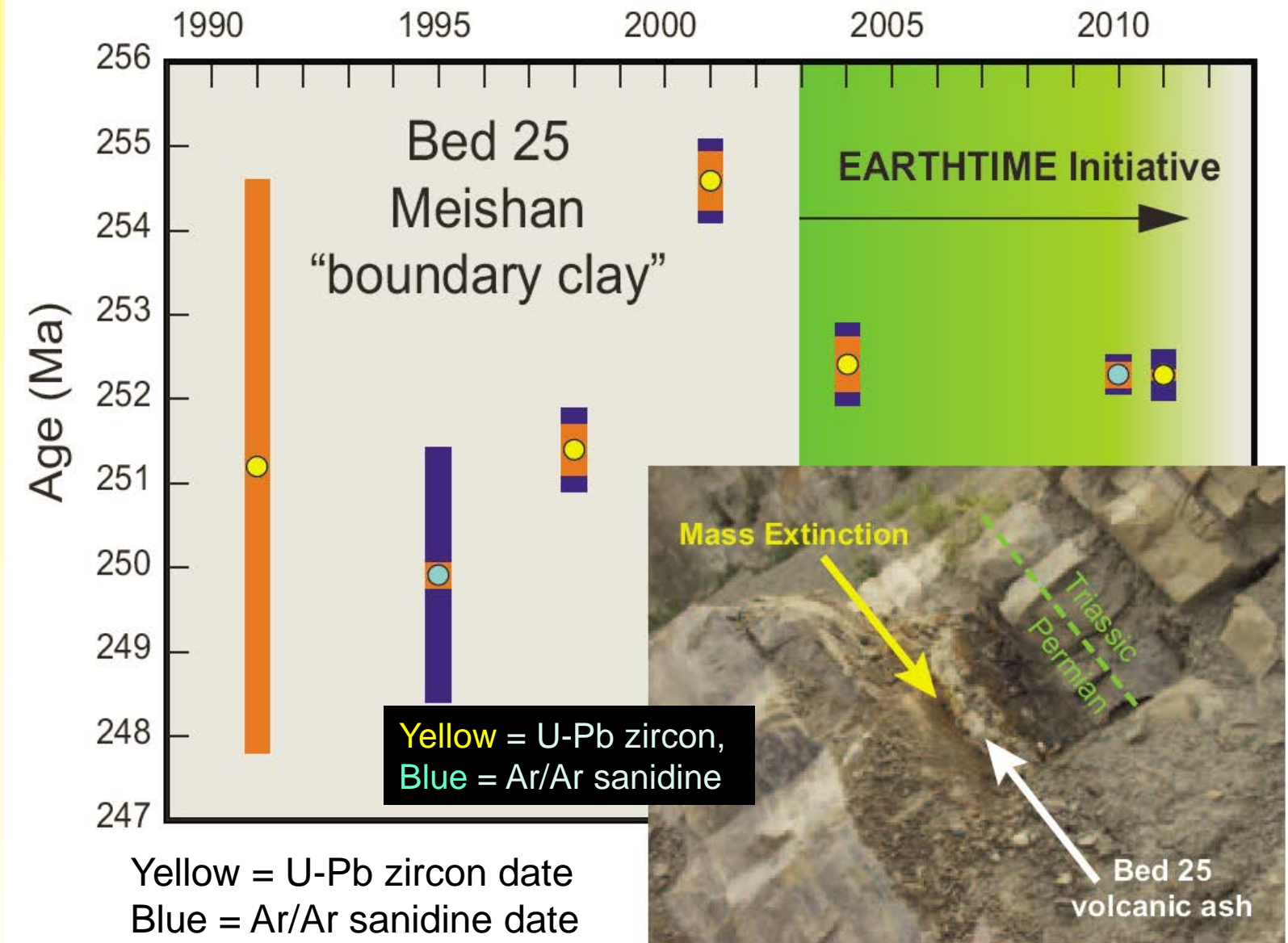
and volcanic eruption in Italy



Permo-Triassic mass extinction

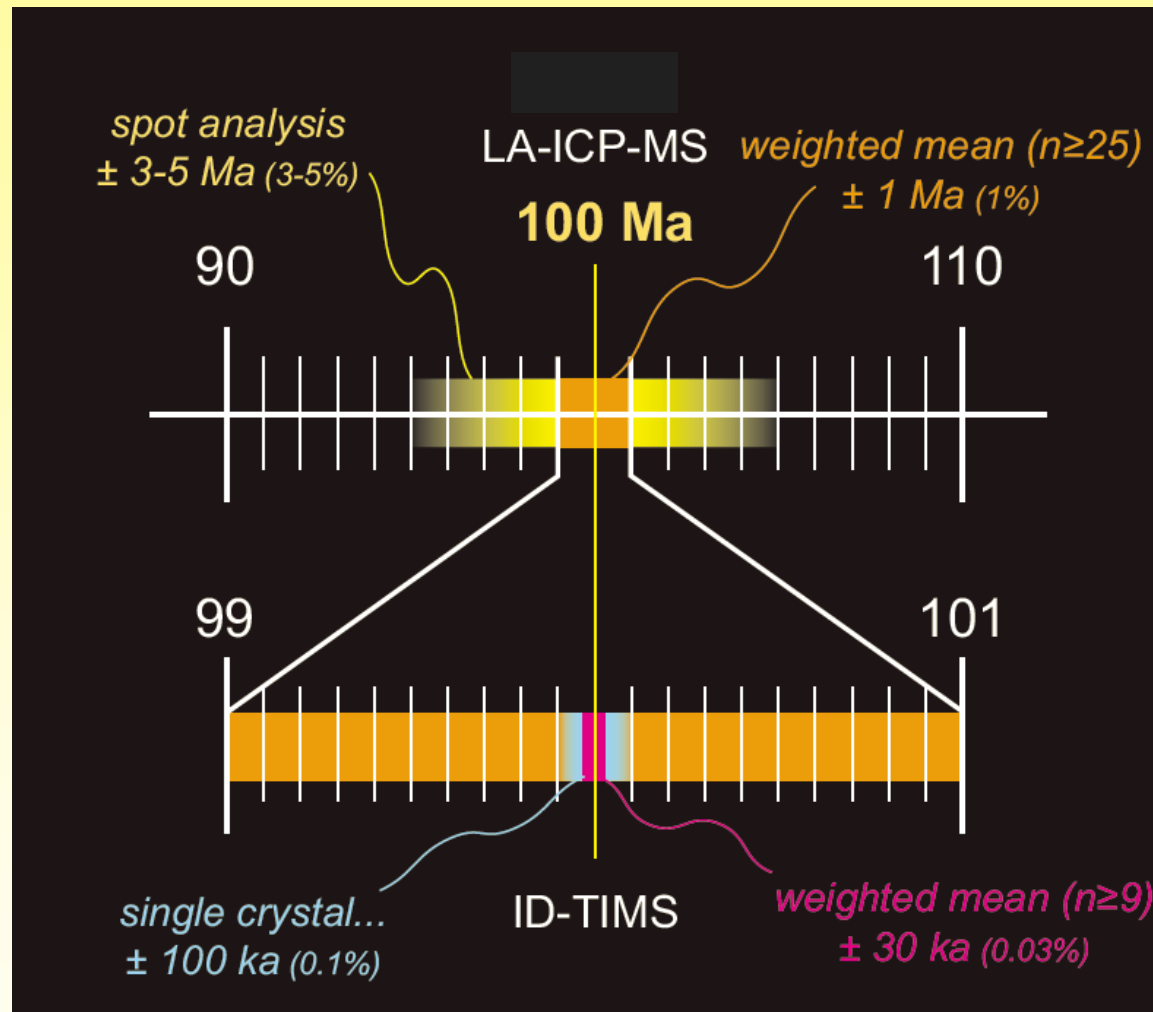


Permo-Triassic mass extinction



Beginning of Siberian Traps volcanism: 250.0 ± 1.6 Ma (Renne et al. 1995)

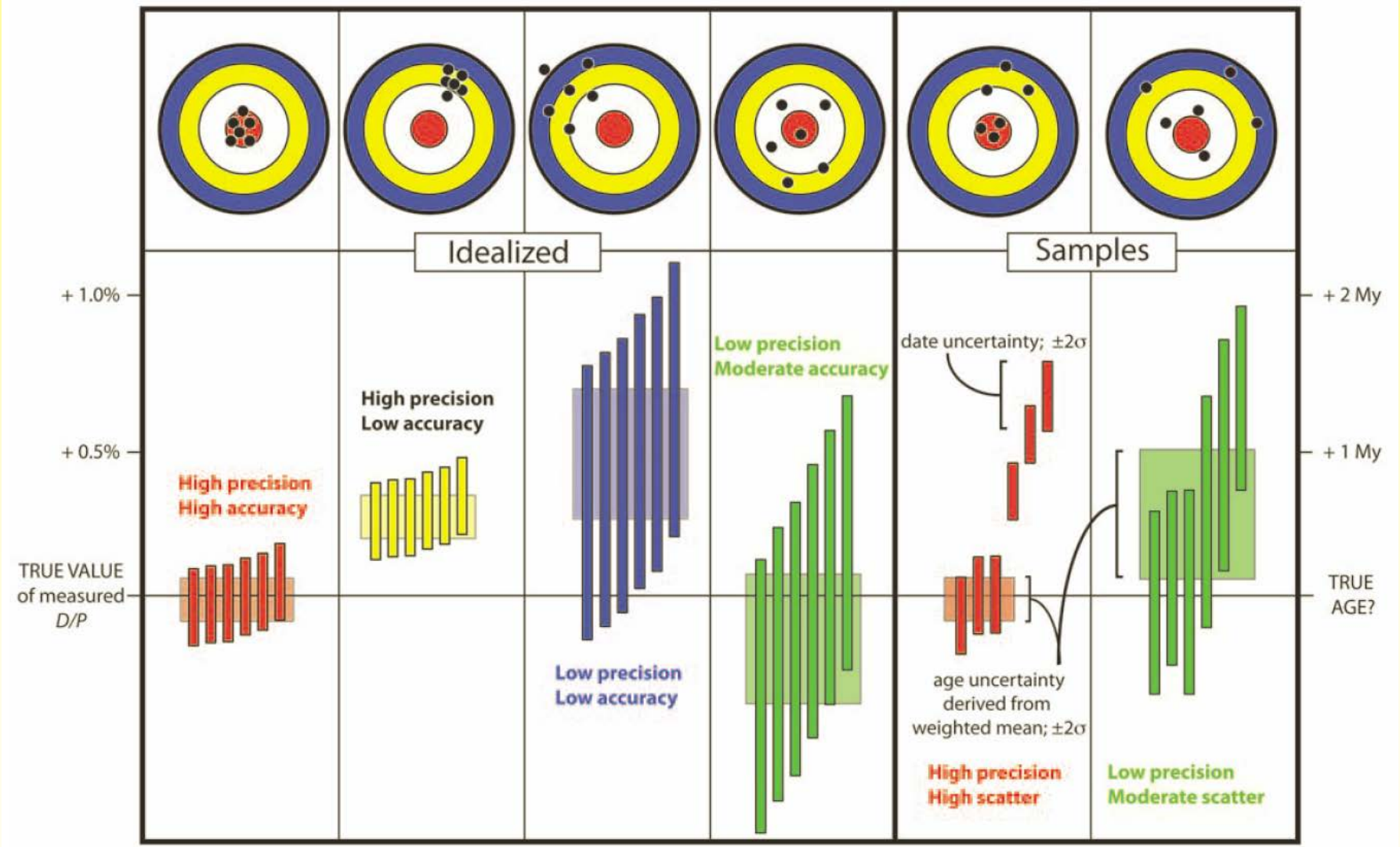
The uncertainty of a data (age) is as important as the data (age) itself
(Ken Ludwig)



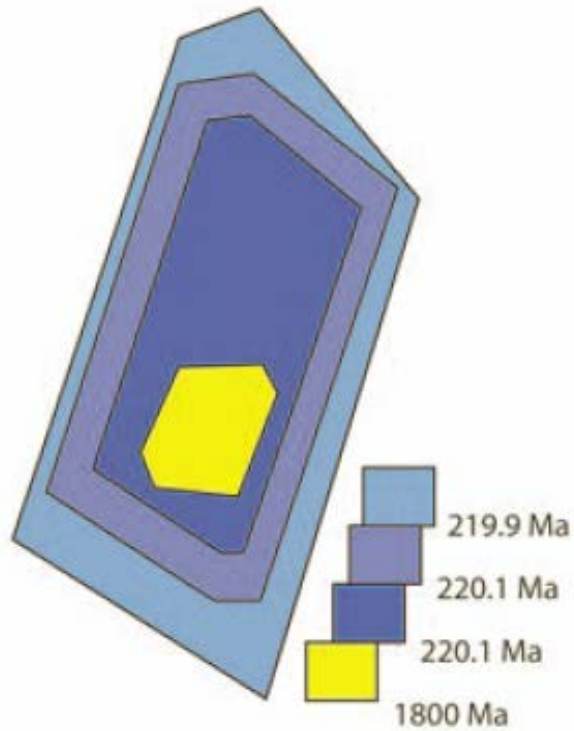
The uncertainty of a date is as important as the date itself (Ken Ludwig 2003)

Ideale Welt

Reale Welt

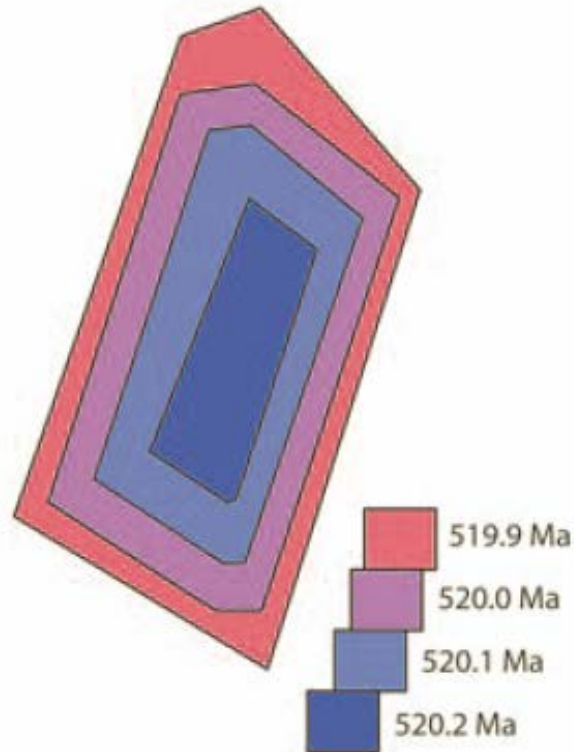


1800 Ma core with a 220 Ma overgrowth



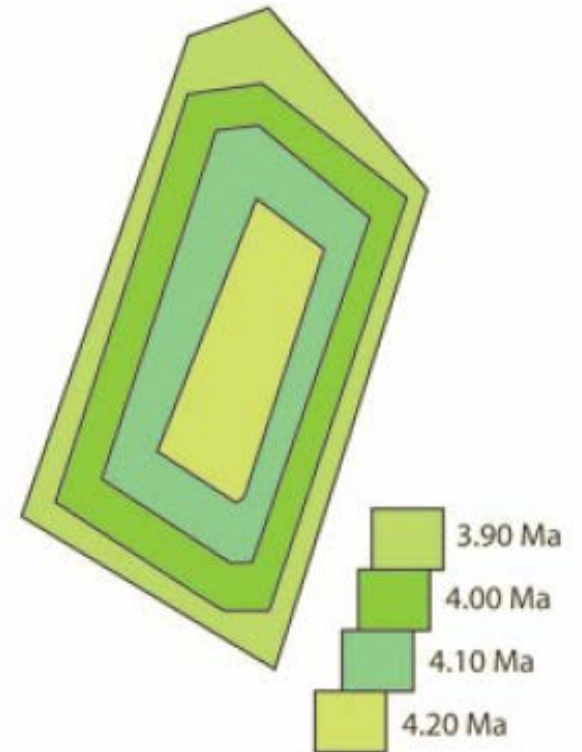
Intracrystal age variation larger than analytical precision of microbeam and ID methods: *In situ* or microsampling required

520 Ma crystal, 300 ky (0.06%) crystallization history



Intracrystal age variation unresolvable: *whole crystal or in situ sample dates are accurate; weighted mean ages may not be*

~4 Ma crystal, 300 ky (7.5%) crystallization history



Intracrystal age variation exceeds analytical precision of microbeam and ID methods: *In situ* or micro-sampling can resolve growth history

Dating methods and Closure temperature

Closure Temp: the temperature at which a cooling mineral can no longer exchange isotopes with its surroundings

Mineral	Method	T (°C)
Zircon	U-Pb	>800
Monazite	U-Pb	>800
Titanite (Sphene)	U-Pb	600
Garnet	Sm-Nd	>550
Hornblende	K-Ar	500
Muscovite	Rb-Sr	500
Muscovite	K-Ar	350
Apatite	U-Pb	350
Biotite	Rb-Sr	300
Biotite	K-Ar	280
K-Feldspar	K-Ar	200
Apatite	Fission Track	120

Closure temperatures for common minerals for different isotopic systems. Note that closure temperatures for different systems in the same minerals can vary.

Datingmethoden und Schließungstemperatur

