¹⁴⁷Sm decays to ¹⁴³Nd by α -decay, decay process: ¹⁴⁷Sm $\rightarrow \alpha$ + ¹⁴³Nd

half-life =106 billion years!!

¹⁴⁷Sm=15% 4 other isotopes

¹⁴³Nd=12.2% 6 other isotopes

De Paolo: *Neodymium Isotope Geochemistry: an introduction* Springer-Verlag 187pp.

A. Dickin: *Nd in the oceans*: <u>http://www.onafarawayday.com/Radiogenic/Ch4/Ch4-5.htm</u>



Sm and Nd are rare earth elements REE have 3+ charge, ionic radii decrease with increasing Z all REE are "incompatible" (they prefer the melt), but light REE are more incompatible



(otherwise purification of Sm and Nd would have been more difficult)

Nd is slightly more incompatible during mantle melting than Sm Sm parent will be enriched in "depleted" sources (i.e. MORB) (opposite to Rb/Sr system, where parent enriched in continents)



(otherwise purification of Sm and Nd would have been more difficult)

High **Sm/Nd** rocks produce more ¹⁴³Nd Low **Sm/Nd** rocks produce less ¹⁴³Nd

Although the difference is small, ¹⁴³Nd/¹⁴⁴Nd increases faster in the mantle than in the crust. Thus, mantle-derived rocks have higher (¹⁴³Nd/¹⁴⁴Nd)₀ than crustal rocks.

Sm/Nd ratios for terrestrial materials:

MORB 0.32 cont. crust ~0.2 seawater 0.211 shale 0.209 garnet 0.539

Based on the decay: $^{147}Sm \rightarrow ^{143}Nd \quad T_{1/2} = 106 \text{ Ga} (\lambda = 6.54 \times 10^{-12} \text{y}^{-1})$

$$\frac{{}^{143}Nd}{{}^{144}Nd} = \left(\frac{{}^{143}Nd}{{}^{144}Nd}\right)_0 + \frac{{}^{147}Sm}{{}^{144}Nd}(e^{\lambda t} - 1)$$

virtually same equation as for Rb/Sr system

Sm-Nd method is useful in Ca-bearing rocks because REE substitute for Ca and garnet (high Sm/Nd ratio)

Sm-Nd method relatively resistant to alteration



Meteorites, i.e. basaltic achondrites Very old rocks from moon and Earth Garnet-bearing metamorphic rocks





Large variations in Sm/Nd ratios in natural rocks are rare Therefore difficulty in obtaining a wide range of Sm/Nd ratios from a single rock body Combined with greater technical demands of Nd-isotope work has limited applications

Applications

- Mineral isochrons for Sm-Nd often work quite successfully because variations in partition coefficients causes moderate variations in Sm-Nd ratios
- Garnet and Cpx have differenet partition coefficients which therefore give rise to large variations in Sm/Nd ratios
- Common occurrence of garnet + cpx is in eclogites where Sm-Nd has been used extensively to date the timing of metamorphism



Sm-Nd remobilisation and re-equilibration

Sm-Nd as REE are relatively immobile and may therefore not fully re-equilibrate during metamorphism



Mineral transformation

- Transformation of igneous augite to metamorphic omphacite
 - Relatively minor cation exchange
 - $(Ca,Mg,Fe,AI)_2(Si,AI)_2O_6$ → $(Na,Ca)(Mg,Fe,AI)Si_2O_6$
 - Monoclinic \rightarrow Monoclinic
 - Often does not completely re-equilibrate
 - Transformation of plagioclase to garnet
 - Major chemical exchange and structural re-organisation
 - CaAl₂Si₂O₈ → Ca₃Al₂Si₃O₁₂
 - Triclinic \rightarrow Isotropic
 - Likely to completely reset Sm-Nd systematics and give the metamorphic age

Effect of LREE-rich inclusions on garnet dating



Price et al. (2000) Chem.Geol. 168

The evolution of Nd isotopes with time in the mantle, the continental crust and the bulk Earth (CHUR)



The evolution of Nd isotopes with time in the mantle, the continental crust and the bulk Earth (CHUR)



$$\varepsilon_{Nd,CHUR} = \begin{bmatrix} \binom{143}{Nd} & \\ Nd & \\ \hline (143}{Nd} & \\ \hline (143}{Nd} & \\ Nd & \\ \hline (143}{Nd} & \\ Nd & \\ CHUR \end{bmatrix} \times 10^4$$

Epsilon notation

$$\varepsilon_{Nd,CHUR} = \left[\frac{\binom{143}{Nd}}{\binom{143}{Nd}}_{A} - 1\right] \times 10^{4}$$

The mantle has a higher ¹⁴⁷Sm/¹⁴⁴Nd ratio than CHUR, so the mantle has been evolving values of ¹⁴³Nd/¹⁴⁴Nd greater than CHUR with time, so $\epsilon_{Nd,CHUR} > 1$ for mantle. The crust has a lower ¹⁴⁷Sm/¹⁴⁴Nd ratio than CHUR, so the crust has been evolving values of ¹⁴³Nd/¹⁴⁴Nd less than CHUR with time, so $\epsilon_{Nd,CHUR} < 1$ for crust.

<u>Model age</u> - a measure of the length of time a sample has been separated from the mantle from which it was originally derived.

- Model ages can be calculated for an individual rock from a single pair of Sm-Nd isotopic ratios.
- The basis of all such model ages is an assumption about the isotopic composition of the mantle source region from which the samples were originally derived.
- Care must be exercised in their interpretation.



$$\left(\frac{{}^{143}Nd}{{}^{144}Nd}\right)_{sample} = \left(\frac{{}^{143}Nd}{{}^{144}Nd}\right)_{initial} + \left(\frac{{}^{147}Sm}{{}^{144}Nd}\right)_{sample} \times (e^{\lambda t} - 1)$$
$$\left(\frac{{}^{143}Nd}{{}^{144}Nd}\right)_{CHUR} = \left(\frac{{}^{143}Nd}{{}^{144}Nd}\right)_{initial} + \left(\frac{{}^{147}Sm}{{}^{144}Nd}\right)_{CHUR} \times (e^{\lambda t} - 1)$$

$$\left(\frac{{}^{143}Nd}{{}^{144}Nd}\right)_{sample} - \left(\frac{{}^{143}Nd}{{}^{144}Nd}\right)_{CHUR} = \left[\left(\frac{{}^{147}Sm}{{}^{144}Nd}\right)_{sample} - \left(\frac{{}^{147}Sm}{{}^{144}Nd}\right)_{CHUR}\right] \times (e^{\lambda t} - 1)$$

$$(e^{\lambda t} - 1) = \left(\frac{({}^{143}Nd / {}^{144}Nd)_{sample} - ({}^{143}Nd / {}^{144}Nd)_{CHUR}}{({}^{147}Sm / {}^{144}Nd)_{sample} - ({}^{147}Sm / {}^{144}Nd)_{CHUR}}\right)$$

$$T_{CHUR} = \frac{1}{\lambda} \ln \left(\frac{({}^{143}Nd / {}^{144} Nd)_{sample} - ({}^{143}Nd / {}^{144} Nd)_{CHUR}}{({}^{147}Sm / {}^{144} Nd)_{sample} - ({}^{147}Sm / {}^{144} Nd)_{CHUR}} + 1 \right)$$







Nd-Sr isotope correlation diagram



Nd-Sr isotope correlation diagram



Summary Sm-Nd



Same timescales as Rb-Sr and K-Ar (Ar-Ar) More resistant to changes during metamorphism and ion exchange Better adherence to closed system assumptions

¹⁴⁶Sm-¹⁴²Nd chronometer

¹⁴⁶Sm–¹⁴²Nd evidence from Isua metamorphosed sediments for early differentiation of the Earth's mantle

Guillaume Caro*, Bernard Bourdon*, Jean-Louis Birck* & Stephen Moorbath†

Nature 423, p428 (2003)



T_{1/2} ¹⁴⁶Sm = 103 Ma Extinct nuclide!

¹⁴²Nd Evidence for Early (>4.53 Ga) Global Differentiation of the Silicate Earth

M. Boyet* and R. W. Carlson

Science 309, p576 (2005)



Nd in the oceans

Nd is not well-mixed in the ocean, because it has a short oceanic residence time (shorter than ocean mixing time) and ratios reflect regional changes in water (or volcanic) inputs.

anything with a residence time shorter than the turnover time of the ocean (~1500 y) will exhibit isotopic variability in seawater

¹⁴³Nd/¹⁴⁴Nd measured in water, skeletal calcium carbonate, heavy-metal sediments, and ferro-manganese nodules to look at changes over time in discrete ocean basins

Nd in the oceans

Nd (and Sr) from different continental sources have isotopic signatures specific to the source rock of the watershed.

Total conc. may be changed during fluvial transport (e.g. Nd = f(pH))



But isotopic ratios ¹⁴³Nd/¹⁴⁴Nd retained

Both Nd and Sr isotopes changed in ocean over time.

The change in ¹⁴³Nd/¹⁴⁴Nd and ⁸⁷Sr/⁸⁶Sr indicate varying contributions from different sources of water to the oceans.

Sr has long oceanic residence time and ratios reflect global changes

Nd has short oceanic residence time (shorter than ocean mixing time) and ratios reflect regional changes in water (or volcanic) inputs.



All rivers plotted drain a variety of continental crusts. Relatively low Nd ratio, relatively high Sr ratio

Table 19.3. Average Weighted Concentrationsand Isotope Ratios of Nd in River Water Priorto Losses due to Sorption in Estuaries

River Water	Discharge, km ³ /y	Nd, ppt	¹⁴³ Nd/ ¹⁴⁴ Nd ^a	
Atlantic Ocean rivers	20,323	55.7	0.511991	4
Pacific Ocean rivers	13,123	27.8	0.512489	•
Indian Ocean rivers	4,878	26.6	0.512191	
Arctic Ocean rivers	4,115	21.6	0.511319	
All rivers	42,439	40.5	0.511330	

Source: Goldstein and Jacobsen, 1987.

^aRelative to 0.512638 for CHUR-Nd.

Atlantic – drains continental crust silicates (low Nd ratio)

 Pacific – drains rocks that were mantle derived volcanics (high Nd ratio)

Nd in the oceans

Comparison of seawater isotope compositions with sea floor Fe-Mn nodules and possible source reservoirs. 0.5120 0.5125



Good isotopic correspondence between river discharge water, seawater, and ferro-manganese nodules

from Alan P. Dickin: http://www.onafarawayday.com/Radiogenic/

Rise of continental crust source to the Atlantic



Rise of volcanic (mantle source) inputs to the Pacific



Some differences due to bottom water inputs...deep circulation.

Short residence time of Nd in the ocean relative to ocean mixing rates allows to obtain basin specific information

Nd in the oceans



Nd in the oceans

