

# Noble gas geochemistry

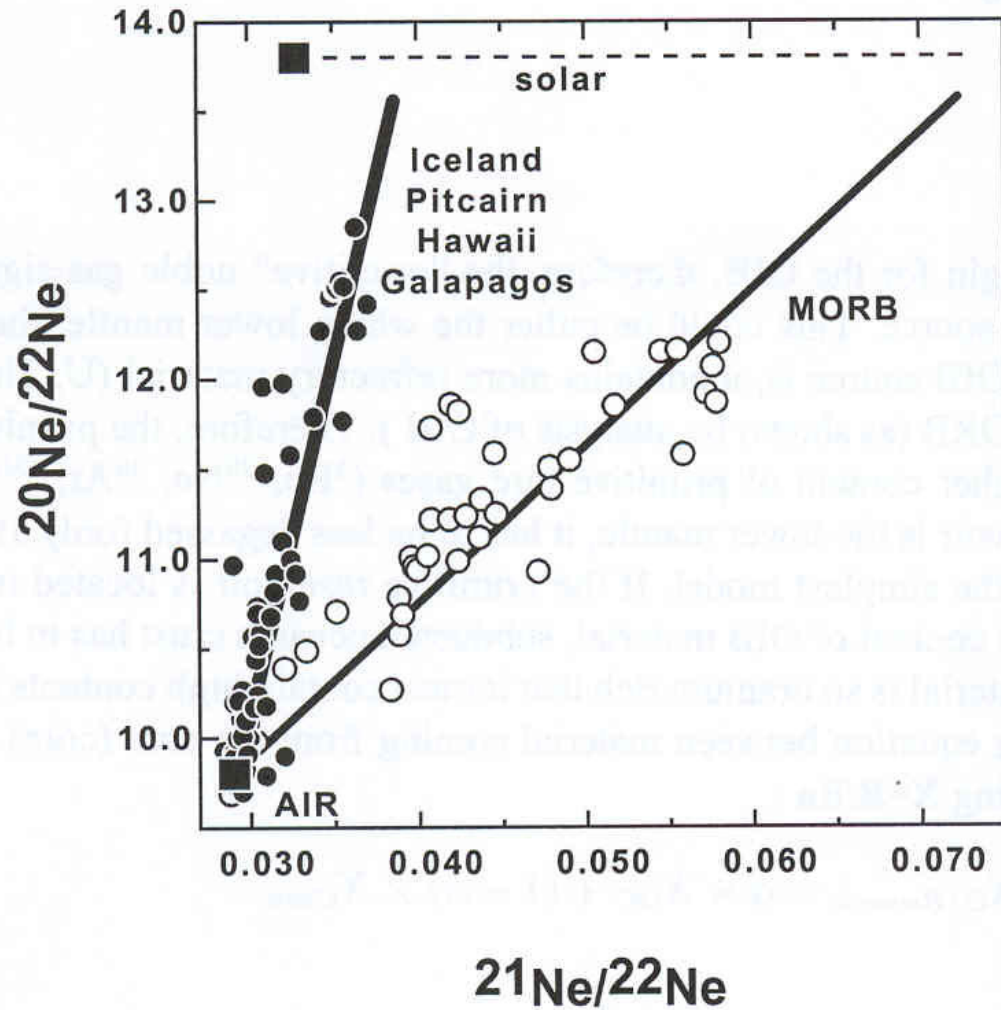
2 4.00260 <b>He</b> Helium	10 20.179 <b>Ne</b> Neon	18 39.948 <b>Ar</b> Argon
36 83.80 <b>Kr</b> Krypton	54 131.30 <b>Xe</b> Xenon	86 (222) <b>Rn</b> Radon

- Primordial noble gases:  $^3\text{He}$ ,  $^{20}\text{Ne}$ ,  $^{36}\text{Ar}$ ,  $^{130}\text{Xe}$
- Produced by radioelement decay:  $^4\text{He}$ ,  $^{21}\text{Ne}$ ,  $^{40}\text{Ar}$ ,  $^{129}\text{Xe}$

# Neon isotopes

- Nobel gas geochemistry

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# Noble gas geochemistry

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$^4\text{He}$ : radiogenic, from  $^{235}\text{U}$ -,  $^{238}\text{U}$ - and  $^{232}\text{Th}$ -decay  
 $^3\text{He}$  cosmogenic or primordial

Upper Mantle depleted in U and Th

Lower Mantle not depleted in U and Th

Upper Mantle  $^3\text{He}/^4\text{He}$ :  $12.6 \times 10^{-6}$

Lower Mantle  $^3\text{He}/^4\text{He}$ :  $\sim 45 \times 10^{-6}$

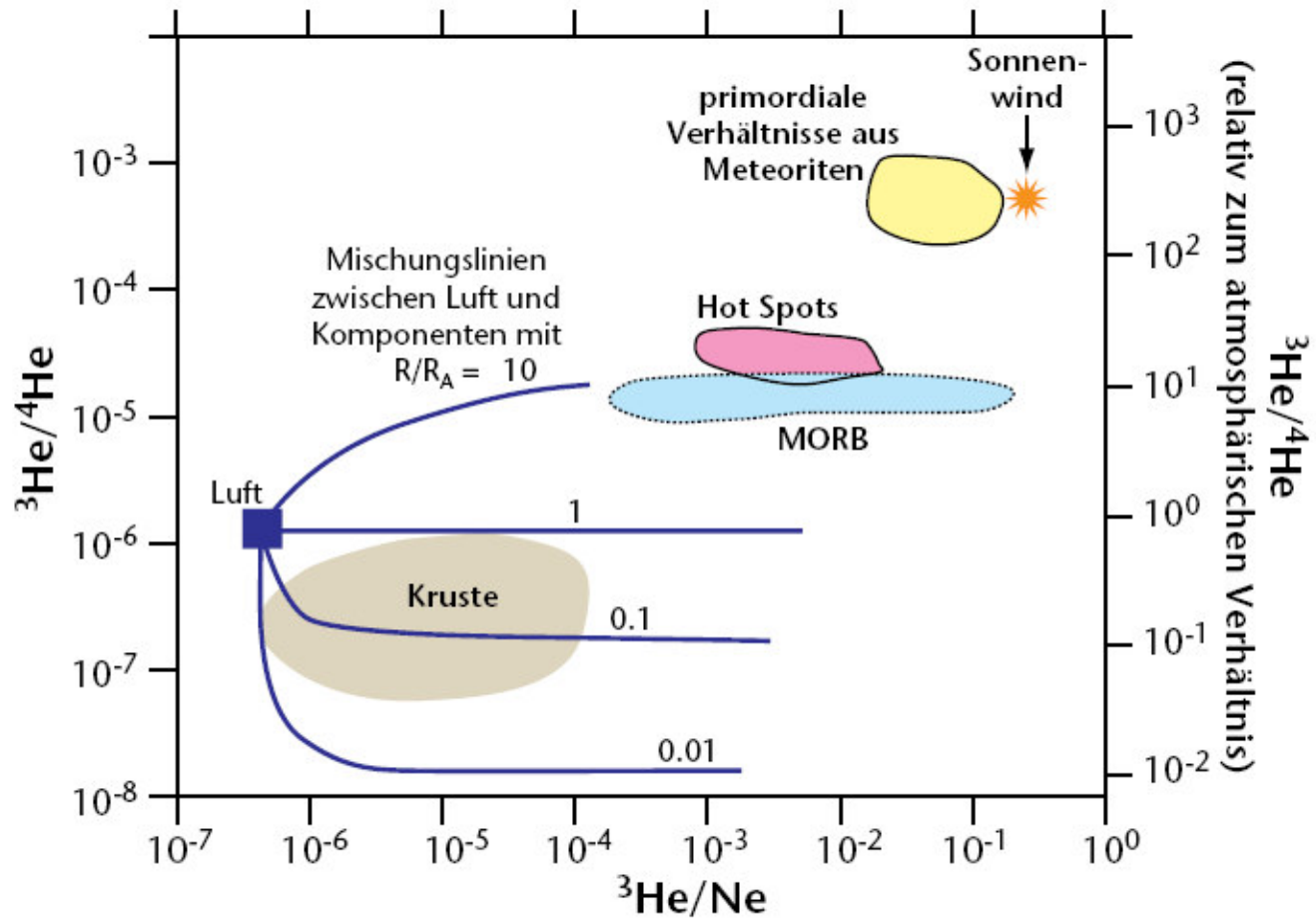
$^3\text{He}/^4\text{He}$  in atmosphere:  $\sim 1.4 \times 10^{-6}$

## Implications: different degrees of degassing

Primordial or undegassed deep mantle reservoir

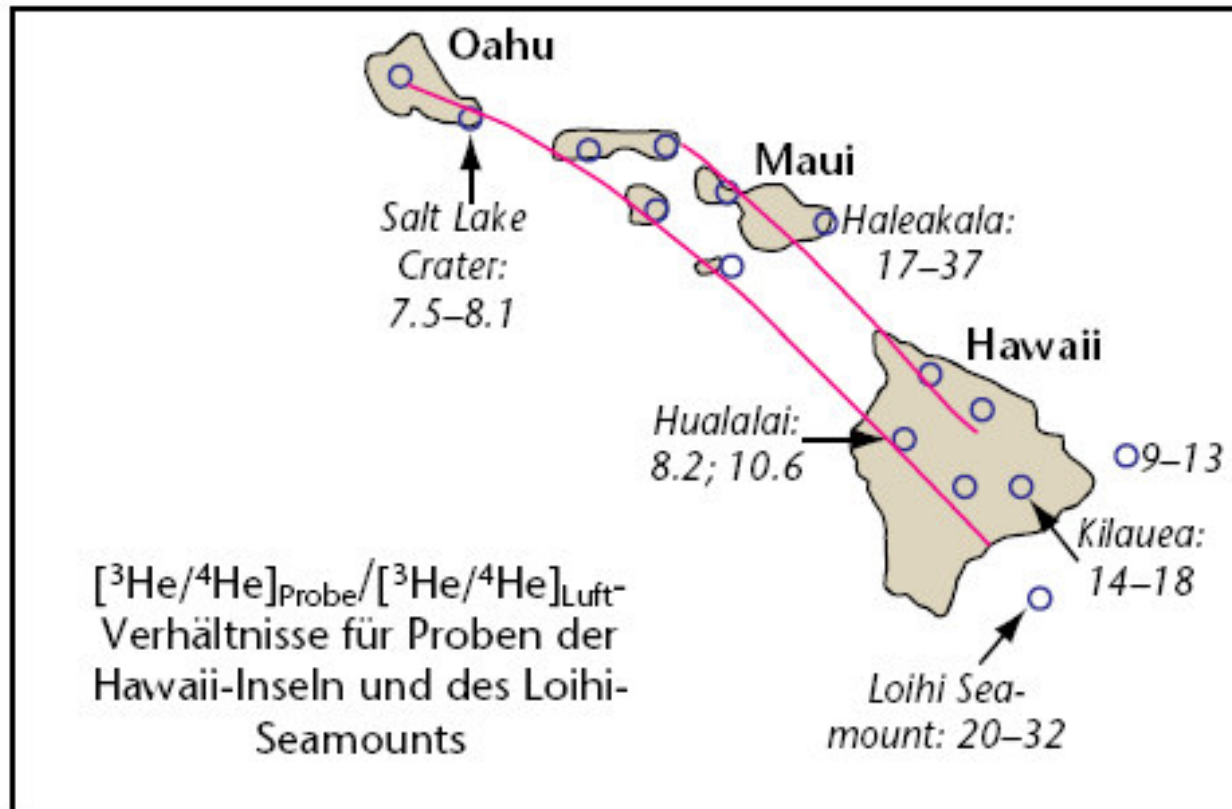
Two layer model of the Earth mantle (?)

# Edelgase



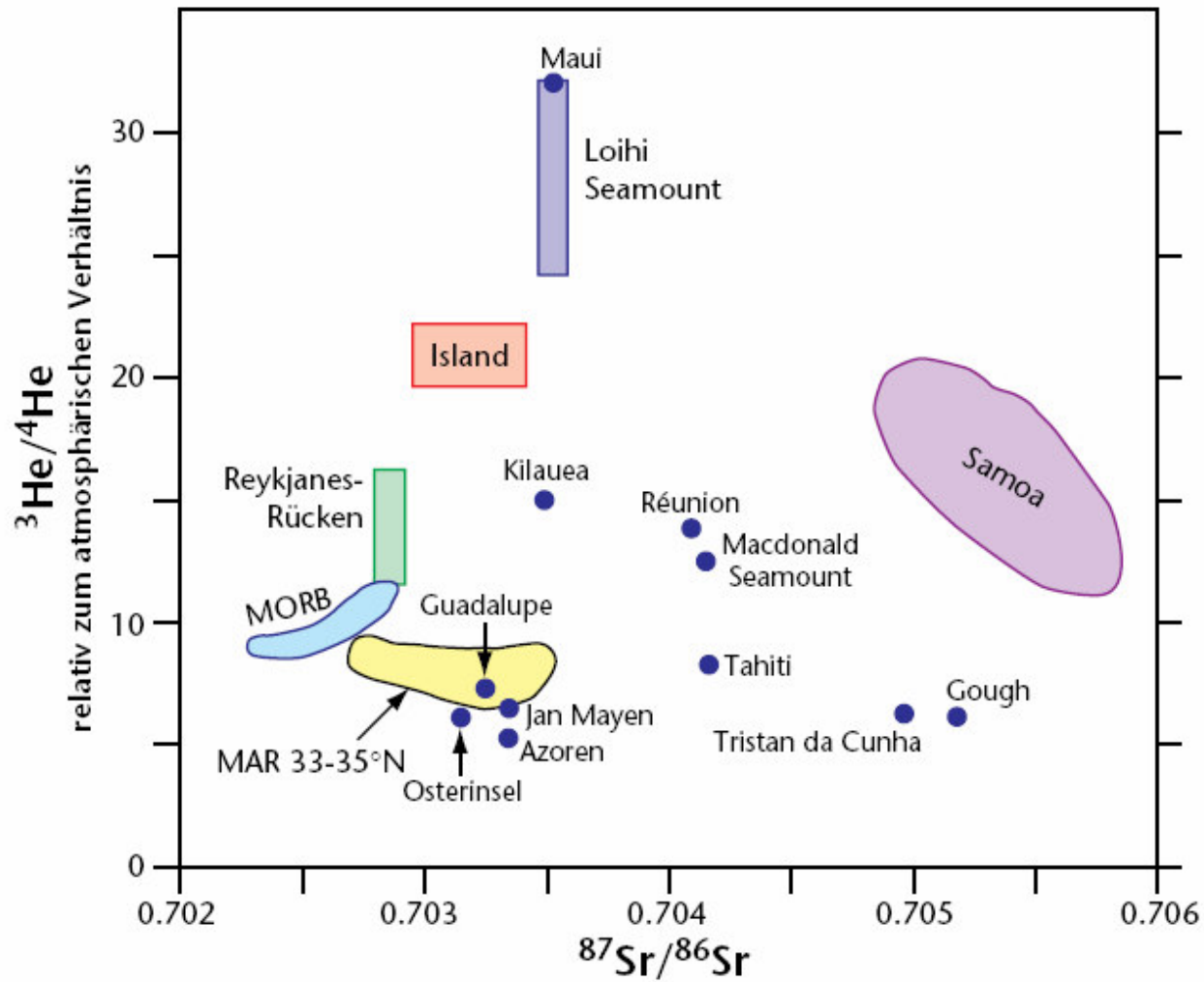
<http://www.dmg-home.de/Ressourcen/Internet-Kurse/Isotopengeochemie.pdf>

# He-Isotopendaten Hawaii



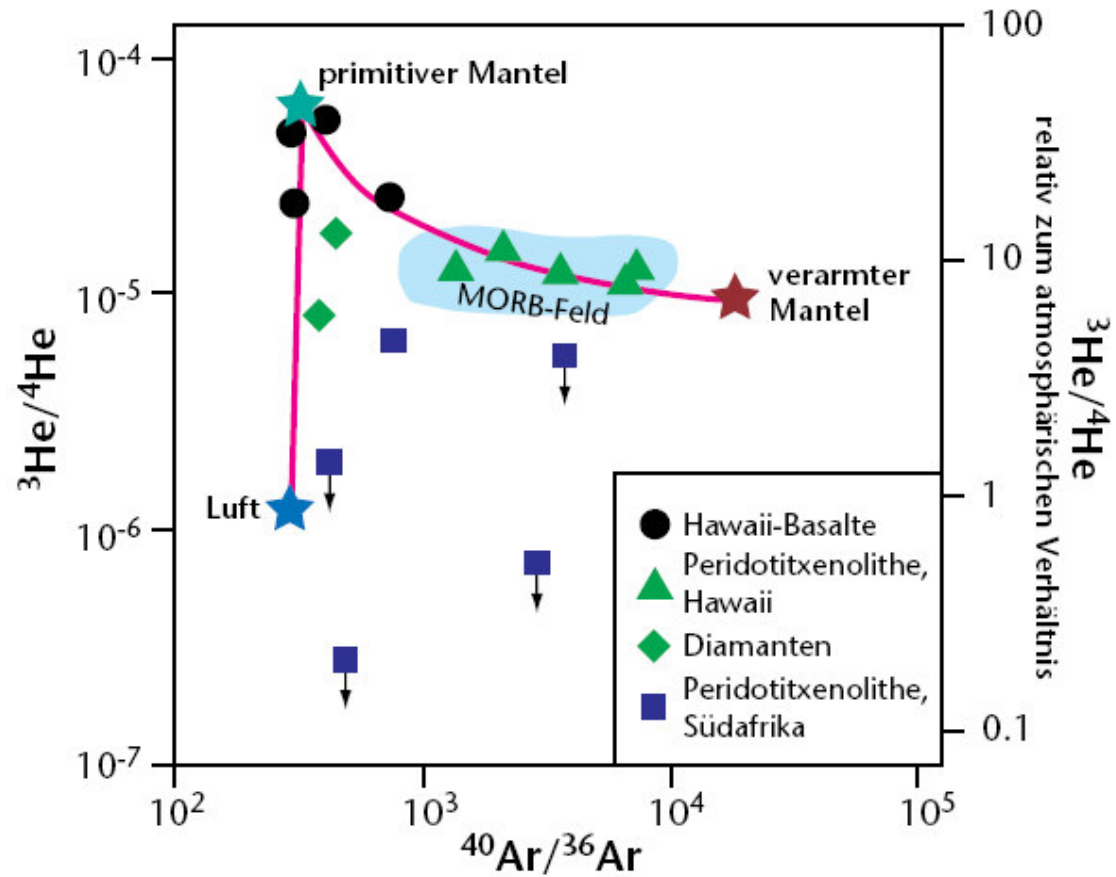
<http://www.dmg-home.de/Ressourcen/Internet-Kurse/Isotopengeochemie.pdf>

# Variation von He und Sr in OIBs und MORBs



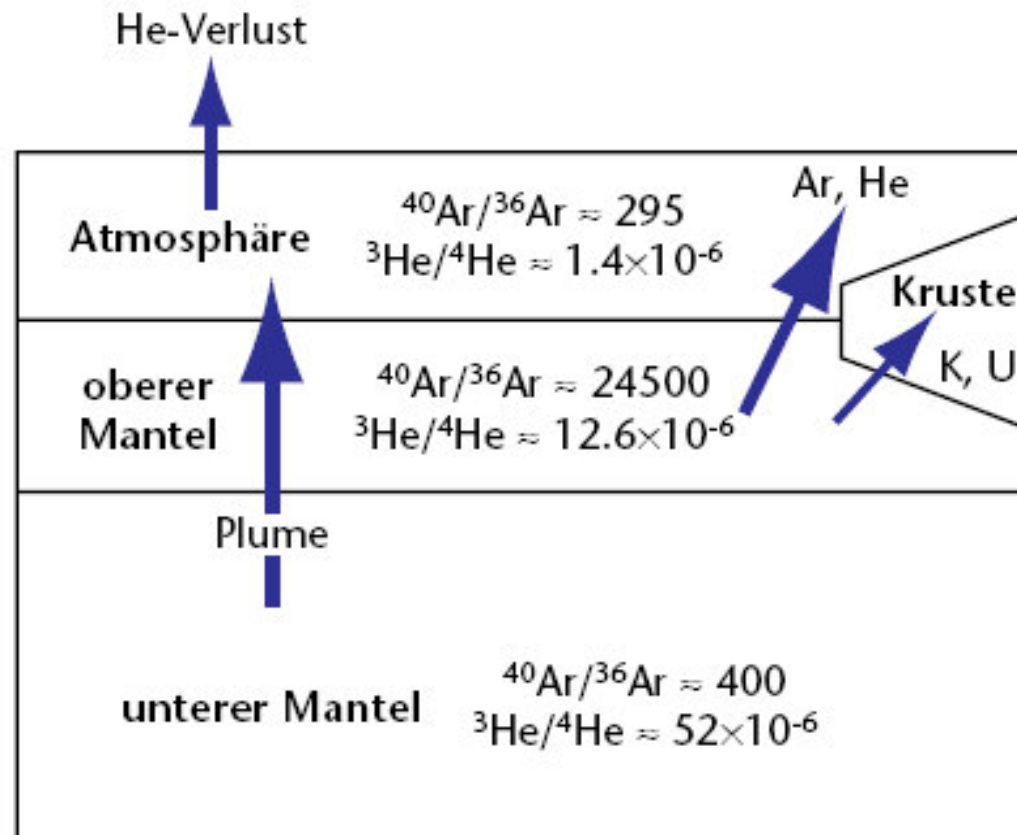
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# He und Ar im Erdmantel



Stosch: Isotopengeochemie

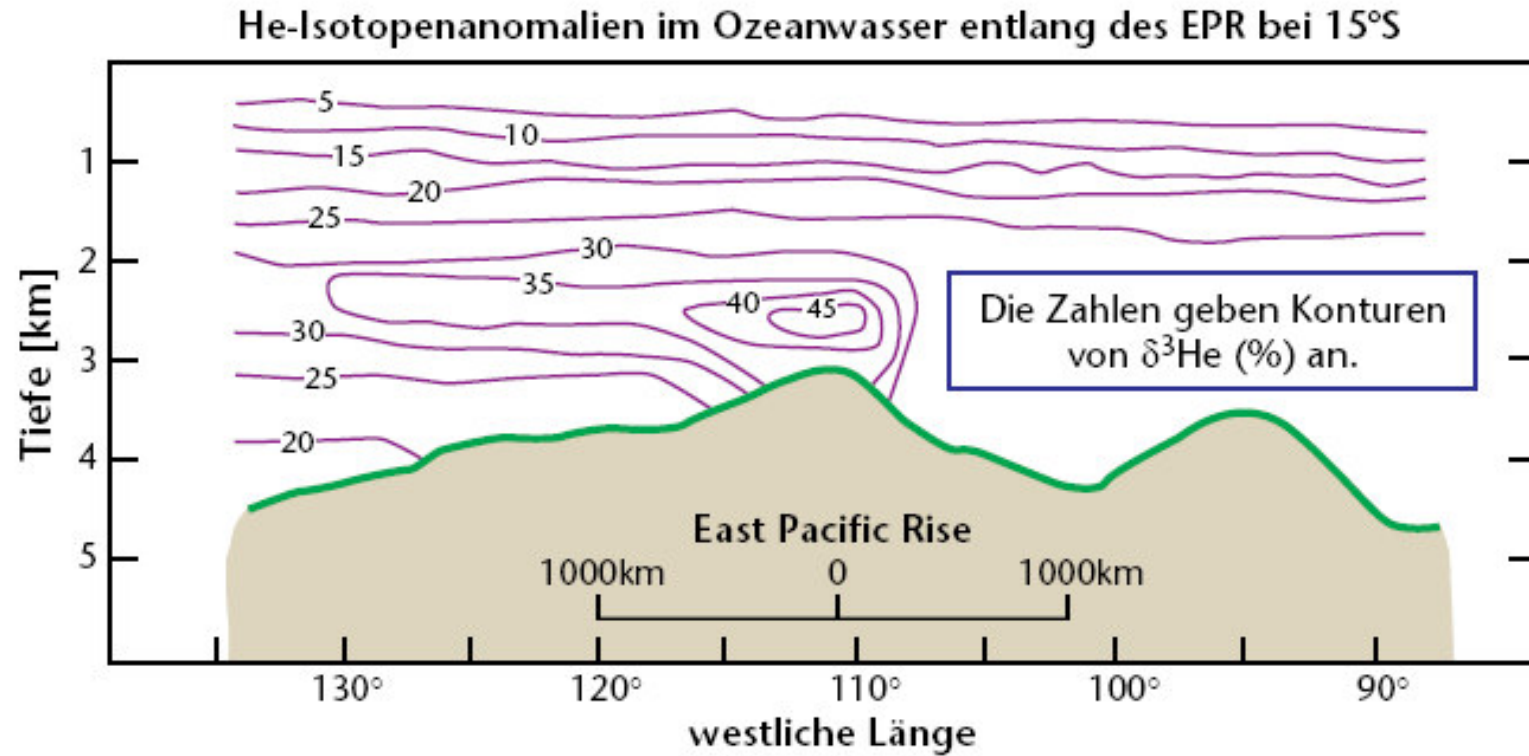
# He und Ar im Erdmantel



Hart 1985 (aus Stosch: Isotopengeochemie)



# He-Entgasung



Stosch: Isotopengeochemie